

Brønsted base (Brönsted base)

A *molecular entity* capable of accepting a *hydron* (proton) from an acid (i.e. a ‘hydron acceptor’) or the corresponding *chemical species*. For example: OH^- , H_2O , CH_3CO_2^- , HSO_4^- , SO_4^{2-} , Cl^- .

See also *conjugate acid–base pair*.

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