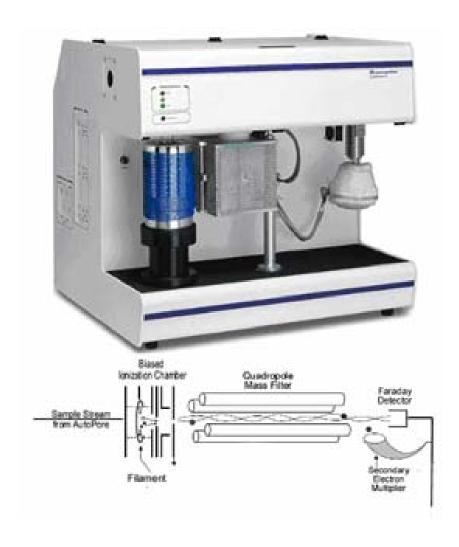
AutoChem[™] II 2920

Bibliography of Peer-Reviewed Papers Citing Use of Micromeritics' AutoChem with a Mass Spectrometer



Citations of AutoChem / Mass Spectrometer Combinations

The list below was compiled from a search of technical papers in peer-reviewed journals. The search criteria were that the paper must cite the use of a Micromeritics AutoChem and also cite use of a mass spectrometer. These conditions do not necessarily mean that the mass spectrometer was connected directly to the AutoChem. Although that information sometimes can be determined from the excerpt (citation) or from the abstract, it most often requires that a full text version of the paper be obtained and the experimental section examined.

	Author /		
Title	Publication	Citation	TPx
A bifunctional catalyst for the single-stage water-gas shift reaction in fuel cell	KG Azzam, IV Babich, K Seshan, L Lefferts - Journal of Catalysis, 2007	Temperature-programmed reduction (H 2 -TPR) studies were conducted in a <u>Micromeritics</u> <u>AutoChem II 2920</u> device. Here, 1 g of catalyst was placed in a U-quartz tube and preheated to 300 °C, then cooled to -75 °C under Ar flow (20 ml/min)	2920
A combination of Ag/alumina and Ag modified ZSM-5 to remove NOx and CO	P Konova, K Arve, F Klingstedt, P Nikolov, A Applied Catalysis B, , 2007	Temperature programmed desorption of octane and propene over the most active catalysts was carried out using a volumetric equipment (<u>AutoChem 2910</u> , <u>Micromeritics</u>) combined with a quadruple <u>mass spectrometer</u> (Carlo Erba Instruments)	2910
A NEW METHOD FOR THE IN-SITU DIFFUSE REFLECTANCE FTIR ANALYSIS OF	6-Mar	minimizing sample size and maximizing throughput is best achieved using an elemental analyzer linked to an isotope ratio <u>mass</u> <u>spectrometer</u> (EA-IRMS). If fuel and thermal NOx can be quantified, this will be an important development to help guarantee combustor	
A novel catalyst of CeO2/Al2O3 for selective catalytic reduction of NO by NH3	Y Shen, S Zhu, T Qiu, S Shen - Catalysis Communications, 2009	Total acidity measurement was evaluated by a temperature programmed desorption (TPD) of ammonia using an AUTOCHEM 2910 (Micromeritics) The ammonia desorption was monitored online by Thermo ONIX ProLab mass spectrometer	2910
A novel route to the preparation of carbon supported nickel phosphide catalysts bydicp.ac.cn	L Ding, M Zheng, A Wang, T Zhang - Catalysis Letters	The gaseous products during the synthesis were analyzed with an in situ <i>mass spectrometer</i> (Oministar, GSD-300). The chemisorption of CO was conducted on a <u>Micromeritics AutoChem II 2910</u> automated catalyst characterization system	2910
A pretreatment method of Ni/[gamma]-Al2O3 catalyst for naphthalene	F Li, X Yi, J Zheng, H Jin, W Fang - Catalysis Communications, 2009	treated with a H 2 /Ar = 10/90 mixture at 300 °C for 15 min, and the contrastive catalyst were carried out at 430 °C for 240 min on the <u>Micromeritics</u> <u>AutoChem II 2920</u> Desorption gases were monitored by an online ThermoStar quadrupole <u>mass spectrometer</u> (model GSD301T2)	2920
A study of uranium oxide based catalysts for the oxidative destruction of short chain	SH Taylor, SR O'Leary - Applied Catalysis B, Environmental, 2000	Hiden quadrupole <u>mass spectrometer</u> . Conversion datalevels using the <u>mass spectrometer</u> . Product selectivitiesobtained using a <u>Micromeritics 2910 AutoChem</u> instrument. Typicallyadsorption using a <u>Micromeritics</u> ASAP 2000 analyser	2910
A study on nanosized cerium oxides systems for environmental catalysisdiva-portal.org [BOOK]	O Adamopoulos - 2003 - kth.diva- portal.org	spectroscopy XANES X-ray absorption near edge structure EXAFS Extended x-ray absorption fine structure BET Brunauer-Emmit-Teller DLS Dynamic light scattering PSD Particle size distribution TEM Transmission electron microscopy MS <u>Mass spectrometer</u> Tign Ignition	

Caps, V., Paraskevas, I., Tsang, S.C., Applied Catalysis A: General, 252 (1), p.37-49, Oct 2003	2 O, CO, O 2, CO 2) were monitored using a <u>mass spectrometer</u> . Its response was sampled every 6 s via a PEQueen's University, Belfast) using an automated <u>Micromeritics AutoChem</u> 2910. The samples were heated at a rate of 10	2910
AJ Dyakonov - Applied Catalysis B, Environmental, 2003	in the 2 ml quartz pulse/flow reactor in an <u>AutoChem 2910</u> catalyst characterization system from <u>Micromeritics</u> and also in a DSC/TG-111 calorimetric–gravimetric setup from Setaram. Reaction gases were analyzed by means of a ThermoStar mass-spectrometer from Balzers	2910
Dyakonov, A.J., Little, C.A., Applied Catalysis B, Environmental, 67 (1), p.52-59, Sep 2006	flow-through reactor (<u>AutoChem 2910</u> , <u>Micromeritics</u>), pretreated incatalytic oxidation The <u>AutoChem 2910</u> catalysis systemwith ThermoStar mass-spectrometer were used to studyby a ThermoStar <u>mass spectrometer</u> from Balzers. The	2910
AJ Dyakonov, CA Little - Applied Catalysis B, Environmental, 2006	Study of catalytic oxidation. The <u>AutoChem 2910</u> catalysis system from <u>Micromeritics</u> coupled with ThermoStar mass-spectrometer were used to study the activity of the CeO 2 /C and Pd/CeO 2 catalysts in a 50 ml/min model gas mixture of 3% CO + 10% O 2 in He	2910
L Kiwi-Minsker, DA Bulushev, A Renken - Journal of Catalysis, 2003	The determination of the concentration of active sites, reactivity, and temperature-programmed (TPD) studies were performed in a <u>Micromeritics</u> <u>AutoChem 2910</u> analyzer provided with a quartz plug-flow reactor. A ThermoStar 200 (Pfeiffer Vacuum) <u>mass spectrometer</u> was used	2910
RV Siriwardane, MS Shen, EP Fisher, JA Poston - Energy Fuels, 2001	The competitive adsorption studies at 25 °C were conducted utilizing <u>Micromeritics</u> <u>AutoChem 2910</u> atmospheric micro reactor The analysis of the outlet gas stream was conducted utilizing a Pfeiffer Vacuum Thermostar <u>mass spectrometer</u>	2910
RV Siriwardane, MS Shen, EP Fisher, J Losch - Energy Fuels, 2005	temperature-programmed desorption (TPD) studies were conducted in a laboratory-scale fixed-bed reactor (Micromeritics model AutoChem 2910 atmospheric The outlet gas stream from the reactor was analyzed using a mass spectrometer (Pfeiffer Vacuum Thermostar)	2910
RV Siriwardane, MS Shen, EP Fisher - Energy Fuels, 2003	Competitive gas adsorption studies were conducted in a lab-scale fixed-bed reactor (Micromeritics AutoChem 2910 atmospheric microreactor) at 14.7 psi (1.01 × 10 The analysis of the outlet gas stream was conducted utilizing a Pfeiffer Vacuum Thermostar mass spectrometer	2910
Diaz, E., Ordonez, S., Vega, A., Journal of Colloid And Interface Science, 305 (1), p.7-16, Jan 2007	previous work[22]. Temperature-programmed desorption studies were carried out in a <u>Micromeritics</u> TPD- <u>2900</u> apparatus connected to a Glaslab 300 <u>mass spectrometer</u> . For this purpose, a 0.50-g carbon sample was heated from 50 to 950 ^oC at 10 ^oC	2900
	Paraskevas, I., Tsang, S.C., Applied Catalysis A: General, 252 (1), p.37-49, Oct 2003 AJ Dyakonov - Applied Catalysis B, Environmental, 2003 Dyakonov, A.J., Little, C.A., Applied Catalysis B, Environmental, 67 (1), p.52-59, Sep 2006 AJ Dyakonov, CA Little - Applied Catalysis B, Environmental, 2006 L Kiwi-Minsker, DA Bulushev, A Renken - Journal of Catalysis, 2003 RV Siriwardane, MS Shen, EP Fisher, JA Poston - Energy Fuels, 2001 RV Siriwardane, MS Shen, EP Fisher, J Losch - Energy Fuels, 2005 RV Siriwardane, MS Shen, EP Fisher, J Losch - Energy Fuels, 2005	Pariaskevas, I., Tsang, S.C., Applied Catalysis A: General, 252 (1), p.37-49, ot 2003 AJ Dyakonov - Applied Catalysis B, Environmental, 2003 AJ Dyakonov, - Applied Catalysis B, Environmental, 2003 Dyakonov, A.J., Little, C.A., Applied Catalysis B, Environmental, 67 (1), p.52-59, Sep 2006 AJ Dyakonov, CA Little - Applied Catalysis B, Environmental, 2006 AJ Dyakonov, CA Little - Applied Catalysis B, Environmental, 2006 AJ Dyakonov, CA Little - Applied Catalysis B, Environmental, 2006 AJ Dyakonov, CA Little - Applied Catalysis B, Environmental, 2006 AJ Dyakonov, CA Little - Applied Catalysis B, Environmental, 2006 AJ Dyakonov, CA Little - Applied Catalysis B, Environmental, 2006 AJ Dyakonov, CA Little - Applied Catalysis B, Environmental, 2006 AJ Dyakonov, CA Little - Applied Catalysis B, Environmental, 2006 AJ Dyakonov, CA Little - Applied Catalysis B, Environmental, 2006 Environmental, 2006 L Kiwi-Minsker, DA Bulushev, A Renken - Journal of Catalysis, 2003 L Kiwi-Minsker, DA Bulushev, A Renken - Journal of Catalysis, 2003 Catalysis, 2003 Catalysis, 2003 Catalysis, 2003 Catalysis, 2003 Environmental, 2006 RV Siriwardane, MS Shen, EP Fisher, JA Poston - Energy Fuels, 2005 RV Siriwardane, MS Shen, EP Fisher, JA Losch - Energy Fuels, 2005 ENVIRONMENT STANDARD S

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Adsorptive removal of tetrahydrothiophene (THT) and tert-butylmercaptan (TBM) using Na-Y and AgNa-Y zeolites for fuel	Lee, D., Ko, E.Y., Lee, H.C., Kim, S., Park, E.D., Applied Catalysis A, General, 334 (1), p.129-136, Jan 2008	in a volumetric unit (ASAP2010, <u>Micromeritics</u>) after treating the samples inmonitoring the effluent using a <u>mass spectrometer</u> (QMS 200, Pfeiffer Vacuum) and <u>AutoChem 2910</u> unit (<u>Micromeritics</u>) equipped with a thermal conductivity	2910
AFeO3 (A=La, Nd, Sm) and LaFe1-xMgxO3 perovskites as methane combustion and CO oxidation catalysts: structural,	Ciambelli, P., Cimino, S., De Rossi, S., Lisi, L., Minelli, G., Porta, P., Russo, G., Applied Catalysis B: Environmental, 29 (4), p.239-250, Feb 2001	Temperature programmed reduction (TPR) experiments were performed using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyzer equipped with a TC detector and coupled with a Hiden HPR 20 <u>mass spectrometer</u> . Samples (100 mg) were preheated in flowing air at 1073 K for 2	2900
Al-and Ga-promoted WO3/ZrO2 strong solid acid catalysts and their catalytic	XR Chen, CL Chen, NP Xu, CY Mou - Catalysis Today, 2004	NH 3 temperature-programmed desorption (NH 3 -TPD) of samples was carried out on a <u>Micromeritics AutoChem 2910</u> instrument The desorption process was monitored by a Quadruple <u>Mass spectrometer</u> (Thermo ONIX, ProLab) connected on line through a heated capillary	2910
Alkylation of phenol with cyclohexanol and cyclohexene using HY and modified HY	R Anand, KU Gore, BS Rao - Catalysis Letters, 2002	the temperature-programmed desorption method, with ammonia as a probe molecule (TPDA), using an Auto- Chem 2910 (Micromeritics , USA) instrument mm 0X25 "m). The GC-MS measurements were performed on a GCMS-QP2000A <u>mass spectrometer</u> equipped with	2910
Alumina supported, perovskite oxide based catalytic materials and their auto-exhaust application	Labhsetwar, N.K., Watanabe, A., Biniwale, R.B., Kumar, R., Mitsuhashi, T., Applied Catalysis B: Environmental, 33 (2), p.165-173, Sep 2001	area, following the standard N-adsorption method, using <u>Micromeritics</u> ASAP-200 instrument. Slurry of perovskite powder was preparedboth thermal conductivity detector (TCD) and quadruple <u>mass spectrometer</u> (Q-Mass). The sample was pre-treated by heating at 800	
AMnO3 (A=La, Nd, Sm) and Sm1-xSrxMnO3 perovskites as combustion catalysts: structural, redox and catalytic	Ciambelli, P., Cimino, S., De Rossi, S., Faticanti, M., Lisi, L., Minelli, G., Pettiti, I., (), Turco, M., Applied Catalysis B: Environmental, 24 (3), p.243-253, Feb 2000	2 were performed using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyser equipped with acoupled with a Hiden HPR 20 <u>mass spectrometer</u> . The sample (30 mg) wasdetected both by TCD and <u>mass spectrometer</u> . TPR profiles of NdMnO 3	2900
Aromatics reduction of pyrolysis gasoline (PyGas) over HY-supported transition metal catalysts	Castano, P., Pawelec, B., Fierro, J.L.G., Arandes, J.M., Bilbao, J., Applied Catalysis A, General, 315, p.101- 113, Nov 2006	obtained using a <u>Micromeritics</u> Digisorb 2600 automaticchemisorption in a <u>Micromeritics</u> ASAP 2010C apparatuswere conducted on a <u>Micromeritics</u> 2900 apparatus providedrecorded using a <u>mass spectrometer</u> quadrupole (Balzers	2900
Assessment of dominant factors affecting liquid phase hydroisomerization on bifunctional zeolites	Funez, A., Thybaut, J.W., Marin, G.B., Sanchez, P., De Lucas, A., Valverde, J.L., Applied Catalysis A, General, 349 (1), p.29-39, Oct 2008	were determined on a <u>Micromeritics</u> ASAP 2010 adsorptiveammonia (TPDA) using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyzer. The sample <u>Micromeritics</u> TPD/TPR <u>2900</u> analyzer. The experimentsSHIMADZU, coupled to a <u>mass spectrometer</u> , QP-5000 SHIMADZU	2900

BIOPROCESSING OF CRUDE OILS	6-Mar	detector (FPD) and a Finnigan ion trap <u>mass</u> <u>spectrometer</u> (ITD) for simultaneous analysespyrolysis-gas chrotpatography- <u>mass</u> <u>spectrometer</u> (Py- GC-MS) utilized a Chemicalperformed with a VG-induced- coupled- <u>mass spectrometer</u> (ICP-MS) .6. Saturate Aromatic	
Brazilian Journal of Chemical Engineering - Influence of thermal treatments on the basic and catalytic properties of [55K]	R. Bastiani, I I. V. Zonno, I I. A. V. Santos, II C. A. Henriques, II, * J, May 2009	adsorption-desorption at 77 K in a <u>Micromeritics</u> ASAP 2000. Before the analysisfor each sample were measured on a <u>Micromeritics</u> <u>2900</u> TPR/TPD analyzer with a quadrupole <u>mass spectrometer</u> detector. Prior to analysis, the	2900
Breaking the dispersion- reducibility dependence in oxide-supported cobalt	A Martínez, G Prieto - Journal of Catalysis, 2007	The reduction behavior of the supported oxidized cobalt phases was studied by hydrogen temperature-programmed reduction (H 2 -TPR) in a Micromeritics AutoChem 2910 device The desorbed hydrogen was monitored in a TCD coupled with a <i>mass spectrometer</i>	2910
Calcination temperature and CuO loading dependence on CuO-CeO2 catalyst activity for water- gas shift reaction	Djinovic, P., Batista, J., Pintar, A., Applied Catalysis A, General, 347 (1), p.23-33, Sep 2008	were performed using a Micromeritics AutoChem II 2920 apparatus onPfeiffer vacuum ThermoStar mass spectrometer, connected to the Micromeritics AutoChem II 2920 apparatusa TCD detector and mass spectrometer. 2.2.3 Catalytic activity	2920
CaO-MgO CATALYSTS FOR SOOT COMBUSTION: KNO3 AS SOURCE FOR DOPING WITH POTASSIUM [118K]	Journal of the Chilean Chemical Society - 3/9/2010	labeled O 2 and secondary ion <u>mass spectrometer</u> (SIMS) showed the formationconventional flow apparatus (<u>Micromeritics</u> Flowsorb 2130) by nitrogenThermostar GSD300T2 quadrupole <u>mass spectrometer</u> QMS200, connected on line	
Carbon dioxide hydrogenation to methanol over the pre-reduced LaCr0. 5Cu0	L Jia, J Gao, W Fang, Q Li - Catalysis Communications, 2009	H 2 -temperature-programmed desorption (H 2 - TPD) was conducted on <u>Micromeritics AutoChem 2920</u> II instrument connected to a ThermoStar GSD 301 T2 <u>mass spectrometer</u> . About 100 mg of catalyst sample was used for each test	2920
Carbon dioxide reforming of methane over La2NiO4 as catalyst precursor—	G Sierra Gallego, F Mondragón, JM Tatibouët, J Catalysis Today, 2008	TPR and H 2 chemisorption experiments were carried out in a <u>Micromeritics AutoChem 2910</u> using about 2 was determined using 30% N 2 /He as the adsorbate on a <u>Micromeritics</u> Flowsorb II 700 °C for 1 h. The reaction products were analyzed by an on-line <u>mass spectrometer</u>	2910
Carbon nanofibers grown on metallic filters as novel catalytic materialsepfl.ch	P Tribolet, L Kiwi- Minsker - Catalysis Today, 2005	Temperature programmed oxidation (TPO) was carried out via an <u>AutoChem 2910</u> instrument (<u>Micromeritics</u> SA, Belgium), connected to a <u>mass spectrometer</u> (Pfeiffer Vacuum) for on-line gas detection. The m/e ratios of 18	2910
Carbon nanotube- supported gold nanoparticles as efficient catalysts for 135.196.210.195	X Tan, W Deng, M Liu, Q Zhang, Y Wang - 135.196.210.195	(Physical Electronics) using Al-Kα radiation. NH3-TPD measurements were performed using a Micromeritics AutoChem II 2920 instrument connected to a ThermoStar GSD 301 T2 mass spectrometer. (3) Catalytic reaction The	2920
Catalytic combustion of gasified biomass over Pt/Al2O3	MFM Zwinkels, GM Eloise Heginuz, BH Applied Catalysis A,, 1997	The Pt dispersion of the fresh catalyst was determined by hydrogenoxygen titration using a <u>Micromeritics</u> TPD TPR <u>2900</u> temperatureprogrammed space velocity was 50,000 h'. All reaction products were analyzed online by a Balzers QMG 421 quadrupole <u>mass</u> <u>spectrometer</u>	2900

Catalytic combustion of hexane over transition metal modified zeolites NaX and CaA	Diaz, E., Ordonez, S., Vega, A., Coca, J., Applied Catalysis B: Environmental, 56 (4), p.313-322, Apr 2005	adsorption at -196 ^oC with a <u>Micromeritics</u> ASAP 2000 surface analyserexperiments were carried out using a <u>Micromeritics</u> TPD- <u>2900</u> apparatus connected to a massusing a Glaslab 300 quadrupole <u>mass spectrometer</u> , which used a capillary inlet	2900
Catalytic combustion of methane on Pd-Cu/SiO2 catalysts	Reyes, P., Figueroa, A., Pecchi, G., Fierro, J.L.G., Catalysis Today, 62 (2), p.209-217, Nov 2000	nitrogen adsorption at 77 K in a <u>Micromeritics</u> Model Gemini 2370. Hydrogenwere carried out in a TPR/TPD <u>2900 Micromeritics</u> system provided with a thermal-conductivitysome experiments a Quadruple <u>Mass spectrometer</u> Hiden HPT 20 was used to detect	2900
Catalytic combustion of methane over LaFeO3 perovskites: the influence of [66K]	Journal of the Chilean Chemical Society - 6/9/2010	BET equation on an automatic <u>Micromeritics</u> apparatus Model ASAP 2010were performed in a TPR/TPD <u>2900 Micromeritics</u> system with a thermal conductivityexperiments, a Quadrupole <u>Mass spectrometer</u> Hiden HPT 20 was used to detect	2900
CATALYTIC COMBUSTION OF TOLUENE ON Pd- Cu/SiO2 CATALYSTS	9-Jun	nitrogen adsorption at 77 K in a <u>Micromeritics</u> Model Gemini 2370 and hydrogencarried out in a TPR/TPD <u>2900 Micromeritics</u> system provided with a thermalsome experiments a Quadrupole <u>Mass</u> <u>spectrometer</u> Hiden HAL 20 was used to detect	2900
Catalytic Conversion of Ethylene to Propylene and Butenes over H- ZSM-5	B Lin, Q Zhang, Y Wang - Industrial & Engineering Chemistry Research	desorption (NH 3 -TPD) and O 2 temperature programmed oxidation (O 2 -TPO) measurements were performed on a <u>Micromeritics AutoChem</u> <u>2920</u> II a rate of 10 K min -1 , and the desorbed NH 3 molecules were detected by ThermoStar GSD 301 T2 <u>mass spectrometer</u> with a	2920
Catalytic decomposition of N2O and catalytic reduction of N2O and N2O+ NO by	A Guzmán-Vargas, G Delahay, B Coq - Applied Catalysis B, , 2003	TPR by H 2 was carried out with a <u>Micromeritics</u> <u>AutoChem 2910</u> apparatus using TCD detection The effluent composition was monitored continuously by sampling on line to a quadruple <u>mass spectrometer</u> (Pfeiffer Omnistar) equipped with Channeltron and Faraday detectors (0	2910
Catalytic decomposition of N2O on supported Pd catalysts: Support and thermal ageing effects on the catalytic	Dacquin, J.P., Dujardin, C., Granger, P., Catalysis Today, 137 (2), p.390-396, Sep 2008	Temperature-programmed reduction experiments (TPR) were performed in a <u>Micromeritics</u> <u>AutoChem II 2920</u> apparatus (5 Vol.% H 2 /Ar, 10 °C/min conductivity detectors for the quantification of NO, O 2, N 2 O and N 2 and a Balzer quadrupole <u>mass spectrometer</u> which allowed	2920
Catalytic deoxygenation of unsaturated renewable feedstocks for production of	M Snåre, I Kubičková, P Mäki- Arvela, D Chichova, K Fuel, 2008	Temperature programmed desorption of hydrogen on the palladium catalyst was performed with the <u>Micromeritics AutoChem 2910</u> coupled to GC–MS (Balzers Instruments, Omnistar) Product identification was validated with a gas chromatograph – <u>mass spectrometer</u> (GC–MS)	2910
Catalytic Hydrodechlorination of Tetrachloroethylene over Pd/TiO2 Minimonoliths	CA González, CM de Correa - 2009	Temperature-programmed oxidation (TPO) in flowing 5% O 2 /Ar was carried out in a Micromeritics AutoChem II 2920 instrument equipped with a thermal conductivity detector (TCD) coupled to a quadrupole mass spectrometer (Pfeiffer Vacuum Omnistar) following the evolution	2920
Catalytic monoliths for ethanol steam reforming	A Casanovas, C de Leitenburg, A Trovarelli, J Llorca - Catalysis Today, 2008	Temperature programmed reduction (TPR) was carried out with a <u>Micromeritics AutoChem II 2920</u> instrument using a H 2 /Ar mixture (5% H 2) at 10 K min –1 and a TCD detector. 2.3 The effluent of the reactor was monitored on line with a MKS Cirrus <u>mass spectrometer</u>	2920

Catalytic partial oxidation of a diesel surrogate fuel using an Ru-substituted	DJ Haynes, A Campos, DA Berry, D Shekhawat, A Roy, Catalysis Today, 2009	Temperature programmed reduction (TPR) and H 2 pulse chemisorption experiments were performed in a <u>Micromeritics AutoChem 2910</u> unit The dry gas products: H 2 , CO, CO 2 , and N 2 were analyzed continuously by an online Thermo Onix <u>mass spectrometer</u> (Model no	2910
Catalytic partial oxidation of n-tetradecane using pyrochlores: Effect of Rh and Sr substitution	Haynes, D.J., Berry, D.A., Shekhawat, D., Spivey, J.J., Catalysis Today, 136 (3), p.206-213, Jul 2008	chemisorption analyses were conducted with a Micromeritics AutoChem 2910 unit. Before the start of the TPR, the catalystcontinuously by means of an online Thermo Onix mass spectrometer (Model no. Prima deltab, a 200a.m.u. scanning	2910
Catalytic Post-Treatment of Automotive Exhaust Gas from Natural Gas Combustion	Y Renème, F Dhainaut, P Granger - Topics in Catalysis	in air at 400 °C during 8 h. H2-temperature-programmed reduction experi- ments (H2-TPR) were performed in a Micromeritics AutoChem II 2920 150 mg with a total flow rate of 30 mL min-1. Reactants and products were analysed using a Balzers QMG 200 <u>mass spectrometer</u>	2920
Catalytic purification of waste gases containing VOC mixtures with Ce/Zr solid solutions	JI Gutiérrez-Ortiz, B de Rivas, R López- Fonseca Applied Catalysis B,, 2006	Temperature-programmed desorption (TPD) of various probe molecules (NH 3 , H 2 O and CO 2) was performed on a <u>Micromeritics AutoChem</u> <u>2910</u> instrument equipped with a thermal conductivity detector (TCD). Prior to adsorption	2910
Catalytic reduction of SO2 with CO over supported iron catalystsdicp.ac.cn	X Wang, A Wang, N Li, X Wang, Z Liu, T Zhang - Ind. Eng. Chem. Res, 2006	NH 3 -TPD of the three acidic supports (HZSM-5, γ-Al 2 O 3 , and SiO 2) and CO 2 -TPD of the MgO support were measured with the <u>Micromeritics</u> <u>AutoChem II 2920</u> Automated Catalyst The CO-TPR spectra were recorded with a <u>mass spectrometer</u> as a detector. 2.5	2920
Ce-Zr-Sr mixed oxide prepared by the reversed microemulsion method for improved Pd-only threeway catalysts	Wang, J., Shen, M., An, Y., Wang, J., Catalysis Communications, 10 (1), p.103-107, Oct 2008	H2-TPR experiments were carried out using a Micromeritics AutoChem 2910. The catalyst was first purged under N2Ar and He were monitored on-line by quadrupole mass spectrometer (Balzers, QMS200). The pulse frequencies were	2910
Changing the Oxygen Mobility in Co/Ceria Catalysts by Ca Incorporation: Implications for Ethanol Steam Reforming	Hua Song and Umit S. Ozkan* J. Phys. Chem. A, October 9, 2009	O2 pulse chemisorption experiments were conducted using <u>AutoChem II 2920 (Micrometrics)</u> connected to a Cirrus <i>Mass Spectrometer</i> (MKS	2920
Characterisation of the deactivation of platinum and palladium supported on activated carbon used as	Ordonez, S., Dez, F.V., Sastre, H., Applied Catalysis B: Environmental, 31 (2), p.113-122, May 2001	Nitrogen porosimetry measurements were performed in a <u>Micromeritics</u> ASAP 2000 apparatus. Morphology and size distributionand reduction (TPR) studies were carried out in a <u>Micromeritics</u> TPD- <u>2900</u> apparatus, equipped with TCD detector, and connected	2900
Characteristics of Fe- exchanged natural zeolites for the decomposition of N2O and its selective catalytic reduction	Ates, A. , Applied Catalysis B, Environmental, 76 (3), p.282-290, Nov 2007	adsorption-desorption (ASAP 2000, Micromeritics) at 77K. Priorperformed with an AutoChem 2910, Micromeritics. In TPD-NH3, thecharacterisation (AutoChem 2910, Micromeriticsusing a quadrupole mass spectrometer (QMS 422 Pfeiffer	2910
Characterization and catalytic performance of vanadium supported on sulfated Ti	J Arfaoui, L Khalfallah Boudali, A Ghorbel, G Journal of Materials 	by tempera- ture-programed desorption (TPD) of ammonia using an <u>AUTOCHEM 2910</u> (<u>Micromeritics</u>) Temperature programed reduction have been carried out in the <u>AUTOCHEM 2910</u> with analyzed by sampling on line with a qua- druple <u>mass spectrometer</u> (Pfeiffer Omnistar	2910

Characterization And Study Of Catalytic Activity Of Cu/Zno/Al2o3 Systems	M Turco, G Bagnasco, C Cammarano, U Costantino, M nt.ntnu.no	TPR measurements were carried out on samples treated in air flow at 450°C using a 5% H2/Ar mixture and a heating rate of 10°C min-1 on a <u>Micromeritics 2900</u> apparatus A <u>mass spectrometer</u> Hiden was employed for identification of products not detected by GC	2900
Characterization of Acid Sites Using Temperature- Programmed Desorptionmicromeritics.co m	TP Desorption - micromeritics.com	use of organic amines and other basic vapors is possible using <u>Micromeritics' AutoChem</u> Series of The temperature zones for the <u>AutoChem</u> should be altered to reflect the use of For the reactive probes (propyl amines), a <u>mass spectrometer</u> is required to quantify the density of	AutoCh em
Characterization of carbon nanofiber composites synthesized by shaping process	P Li, TJ Zhao, JH Zhou, ZJ Sui, YC Dai, WK Yuan - Carbon, 2005	The TPD profiles were examined on a Micromeritics AutoChem II apparatus rate of the furnace (10 °C/min) were controlled and the amounts of CO and CO 2 desorbed from the carbon materials (about 0.25 g) were monitored with an ABB Questor GP process mass spectrometer	2920
Characterization of coke deposited on Pt/alumina catalyst during reforming of liquid hydrocarbons	Shamsi, A., Baltrus, J.P., Spivey, J.J., Applied Catalysis A, General, 293, p.145- 152, Sep 2005	were measured. A <u>Micromeritics</u> Pulse Chemisorbcarried out by using a <u>Micromeritics</u> <u>AutoChem II</u> instrument. Thethe <u>Micromeritics</u> <u>AutoChem</u> instrument. 2.2by a quadrupole <u>mass spectrometer</u> using internal and	2920
Characterization of precursors and reactivity of LaNi1- xCoxO3 for the partial	GC de Araujo, S Lima, MC Rangel, VL Parola, MA Catalysis Today, 2005	2000 apparatus. Prior to the adsorption experiments, the samples were outgassed at 423 K for 2 h. Temperature-programmed reduction profiles were taken with a <u>Micromeritics</u> TPD/TPR <u>2900</u> apparatus. The sample (ca. 30	2900
Characterization of surface oxygen complexes on carbon nanofibers by TPD, XPS	JH Zhou, ZJ Sui, J Zhu, P Li, D Chen, YC Dai, WK Yuan - Carbon, 2007	designations. 2.2. TPD. The TPD experiments were carried out on AutoChem II 2920 (Micromeritics , USA). A min. The downstream gas was diverted to a quadrupole <u>mass</u> <u>spectrometer</u> (Questor, ABB Extrel, USA) for analysis	2920
Characterizations of Unsupported and Supported Rhodium- Iron Phosphate	Y Wang, Q Yuan, Q Zhang, W Deng - 2007	Characterization Method. Temperature-programmed reduction with CO (CO TPR) was performed on an <u>AutoChem 2920</u> II instrument (<u>Micromeritics</u> Instrument Co.) connected to a ThermoStar <u>mass spectrometer</u> (Pfeiffer Vacuum)	2920
Co 3 O 4/CoAl 2 O 4 🖶	정진혁, 박종원,	│ │ 본 실험에서 사용한 반응장치는 제조된	
이용한 화학적 순환 연소	ㅇᆫㄱ, ㄱㅇᆫ, 주윤경, 박종수,	ᆫ ᆯᆷ에서 겡ᆫ ᆫᆼᆼ진 제ㅗ ᆫ 금속산화물 매체의 기초 산화 환원 활성을 분석하기	
.,02 -, 1 122 22	정헌, 이호태, 윤 cheric.org	위해 U-type 미분 반응기(<u>Micromeritics</u> Co.	
		AutoChem 온도는 20°C/min의 승온 속도로	AutoCh em
		50℃에서 1000℃까지 온도 조건하에서 반응 실험을	
		실시하여 TCD와 <u>mass spectrometer</u> 로 분석하였다. 	
CO oxidation and methane combustion on LaAl1-xFexO3 perovskite solid	P Ciambelli, S Cimino, G Lasorella, L Lisi, S De Applied Catalysis B, , 2002	Temperature programmed reduction (TPR) experiments were performed using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyser equipped with a TC detector and coupled with a Hiden HPR 20 <u>mass</u> <u>spectrometer</u> . Samples (0.1 g) were	2900

CO2 reforming of CH4 over La–Ni based perovskite precursors	GS Gallego, F Mondragón, J Barrault, JM Applied Catalysis A, , 2006	TPR and H 2 chemisorption experiments were carried out in a <u>Micromeritics AutoChem 2910</u> using about 2 was measured using 30% N 2 /Ar as the adsorbate on a <u>Micromeritics</u> Flowsorb II at 700 °C for 1 h. The reaction products were analyzed by an online <u>mass spectrometer</u>	2910
Cobalt and Copper Composite Oxides as Efficient Catalysts for Preferential	D Li, X Liu, Q Zhang, Y Wang, H Wan - Catalysis Letters, 2009	lysts were studied by temperature-programmed reduction techniquesincluding H 2 -TPR and CO-TPR with a <u>Micromeritics</u> Auto Chem <u>2920</u> II detector (TCD), and the amount of CO consumption was measured using a Pfeiffer Vacuum ThermoStar <u>mass spectrometer</u> with a	2920
Cobalt supported on morphologically tailored SBA-15 mesostructures: The impact of pore length on metal dispersion and	Prieto, G., Martinez, A., Murciano, R., Arribas, M.A., Applied Catalysis A, General, 367 (1), p.146-156, Oct 2009	were recorded in a <u>Micromeritics</u> ASAP 2000 deviceH2 -TPR) in a <u>Micromeritics</u> <u>AutoChem</u> 2910 device. Thedownstream and a <u>Mass</u> <u>spectrometer</u> on-line registeredin an ASAP 2010C <u>Micromeritics</u> equipment by extrapolating	2910
Combustion of trichloroethylene and dichloromethane over protonic zeolites: Influence of adsorption properties on the	Intriago, L., Diaz, E., Ordonez, S., Vega, A., Microporous and Mesoporous Materials, 91 (1), p.161-169, Apr 2006	adsorption at -196^oC with a Micromeritics ASAP 2000 surface analyserstudies were carried out using a Micromeritics TPD-2900 apparatus connected to a massusing a Glaslab 300 quadrupole mass spectrometer, which used a capillary inlet	2900
Compact string reactor for autothermal hydrogen productionepfl.ch	C Horny, A Renken, L Kiwi-Minsker - Catalysis Today, 2007	Temperature-programmed reduction (TPR) and oxidation (TPO) were carried out in a <u>Micromeritics AutoChem 2910</u> apparatus by passing, respectively The outlet concentrations were monitored by a <u>mass spectrometer</u> (Thermostar 200, Pfeiffer Vacuum) to obtain TPO and TPR	2910
Comparative study of CuO- CeO2 catalysts prepared by wet impregnation and deposition-precipitation	Gurbani, A., Ayastuy, J.L., Gonzalez-Marcos, M.P., Herrero, J.E., Guil, J.M., Gutierrez- Ortiz, M.A., International Journal of Hydrogen Energy, 34 (1), p.547-553, Jan 2009	N2adsorption-desorption isotherms at 78K (<u>Micromeritics</u> ASAP 2010). The crystallineCO-TPR was followed by a <u>mass spectrometer</u> (MS) (MKS Cirrus 300) coupledsame experimental equipment (<u>Micromeritics</u> <u>AutoChem</u> <u>2910</u>) with about 0.4g of	2910
Comparative study of CuO– CeO2 catalysts prepared by wet impregnation and	A Gurbani, JL Ayastuy, MP González-Marcos, International Journal of, 2009	was continuously monitorized by TCD detector; CO consumption in CO-TPR was followed by a <u>mass spectrometer</u> (MS) (MKS TPR, TPD and OSC measurements were conducted in the same experimental equipment (<u>Micromeritics</u> <u>AutoChem 2910</u>) with about 0.4 g of sample	2910
Comparison of adsorption properties of a chemically activated and a steamactivated carbon, using inverse gas	Di@?az, E., Ordonez, S., Vega, A., Coca, J., Microporous and Mesoporous Materials, 82 (1), p.173-181, Jul 2005	by nitrogen adsorption at -196^oC (Micromeritics ASAP 2000 surface analyser), assumingdesorption studies were carried out in a Micromeritics TPD-2900 apparatus connected to a Glaslab 300 mass spectrometer. For this purpose, 0.50g activated	2900
Comparison of the promoting effects of gallium and aluminium on the n-butane	CJ Cao, XZ Yu, CL Chen, NP Xu, YR - Reaction Kinetics and, 2004	NH 3 -TPD and H 2 -TPR were carried out using a Micromeritics AutoChem 2910 instrument. A 0.1 g sample was used for each TPD or TPR experiment The NH 3 -TPD and H 2 - TPR processes were monitored with a Quadruple Mass spectrometer (Thermo ONIX, ProLab)	2910

Comparison of water-gas shift reaction activity and long-term stability of nanostructured CuO-CeO2 catalysts prepared	Djinovic, P., Batista, J., Levec, J., Pintar, A., Applied Catalysis A, General, 364 (1), p.156-165, Jul 2009	measurements were performed using a Micromeritics ASAP 2020 MP/C apparatusmeasurements were performed on a Micromeritics AutoChem II 2920 catalyst characterizationidentified and quantified using a mass spectrometer (Pfeiffer Vacuum, model Thermostar	2920
Comparison of water–gas shift reaction activity and long-term stability of	P Djinović, J Batista, J Levec, A Pintar - Applied Catalysis A, General, 2009	H 2 -TPR/TPD measurements were performed on a <u>Micromeritics AutoChem II 2920</u> catalyst characterization system, using 250 mg of a sample The desorbed gasses were identified and quantified using a <u>mass spectrometer</u> (Pfeiffer Vacuum, model Thermostar). 2.5	2920
Condensation of glyceraldehyde acetonide and acetone over basic catalysts	Veloso, C.O., Henriques, C.A., Dias, A.G., Monteiro, J.L.F., Catalysis Today, 107, p.294- 301, Oct 2005	N2adsorption-desorption at 77K in a <u>Micromeritics</u> ASAP 2000. Before the analysis, thedesorption profile were measured on a <u>Micromeritics</u> 2900 TPR/TPD analyzer with a quadrupole <u>mass spectrometer</u> detector (Balzers QMS-200). 2.4 Reaction	2900
Condensation of glyceraldehyde acetonide with ethyl acetoacetate over Mg, Al	CO Veloso, CN Pérez, BM de Souza, EC Lima, - Microporous and , 2008	TPD analyses were run under He at a heating rate of 20 °C min –1 up to 450 °C. The amount of CO 2 chemisorbed and its desorption profile were measured on a <u>Micromeritics 2900 TPR/TPD</u> analyzer with a quadrupole <u>mass spectrometer</u> detector (Balzers QMS-200). 2.4	2900
Conversion of Cellulose into Sorbitol over Carbon Nanotube-Supported	W Deng, X Tan, W Fang, Q Zhang, Y Wang - Catalysis Letters	NH3 temperature-programmed desorption (NH3-TPD) and H2 temperature-programmed desorption (H2-TPD) were performed on a <u>Micromeritics</u> <u>AutoChem 2920 II</u> rate of 10 °C min-1. The desorbed NH3 were monitored by ThermoStar GSD 301 T2 <u>mass spectrometer</u> with	2920
Copper-based efficient catalysts for propylene epoxidation by molecular oxygen	Y Wang, H Chu, W Zhu, Q Zhang - Catalysis Today, 2008	H 2 -TPR and NH 3 -TPD were performed using a Micromeritics AutoChem II 2920 instrument TPD was performed in He flow by raising the temperature to 973 K at a rate of 10 K min -1. The desorbed NH 3 was detected with a mass spectrometer (ThermoStar GSD 301 T2) by	2920
Cu/±c-Al2O3 catalyst for the combustion of methane in a fluidized bed reactor	lamarino, M., Chirone, R., Lisi, L., Pirone, R., Salatino, P., Russo, G., Catalysis Today, 75 (1), p.317-324, Jul 2002	Temperature programmed reduction (TPR) experiments were performed using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyser equipped with a TC detector and coupled with a Hiden HPR 20 <u>mass spectrometer</u> . Catalyst (100 mg) was preheated in air at 600 C for 2 h and, after	2900
Cu/ZnO/Al2O3 catalysts for oxidative steam reforming of methanol: The role of Cu	M Turco, G Bagnasco, C Cammarano, P Applied Catalysis B, , 2007	N 2 O passivation method that involves the formation of surface Cu 2 O and subsequent H 2 reduction [47], using the same Micromeritics 2900 apparatus allowed analysis of H 2 , CO (detection limit = 0.01%), CO 2 , O 2 , CH 4 , CH 3 OH, H 2 O. A <u>mass spectrometer</u> Hiden was	2900
Cu/γ-Al2O3 catalyst for the combustion of methane in a fluidized bed reactor	M Iamarino, R Chirone, L Lisi, R Pirone, P Salatino, G Catalysis Today, 2002	analysis. Temperature programmed reduction (TPR) experiments were performed using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyser equipped with a TC detector and coupled with a Hiden HPR 20 <u>mass spectrometer</u> . Catalyst (100	2900

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Deactivation due to sulfur poisoning and carbon deposition on Rh-Ni/Al2O3 catalyst during steam reforming of	SL Lakhapatri, MA Abraham - Applied Catalysis A, General, 2009	Temperature programmed oxidation (TPO) and reduction (TPR) studies were done using a Micromeritics AutoChem 2910 equipped with a thermal conductivity using 3% O 2 . For TPO of the used catalyst, the instrument was coupled with a mass spectrometer to measure the	2910
Deactivation Kinetics for Direct Dimethyl Ether Synthesis on a CuO- ZnO- Al2O3/γ	I Sierra, J Ereña, AT Aguayo, M Olazar, J Bilbao - 2009	The metal surface is 11.7 m 2 (g of catalyst) -1 and has been determined by N 2 O chemisorption (Micromeritics AUTOCHEM 2920 online with a Balzers Instruments Omnistar mass spectrometer). The physical properties	2920
Deactivation of a CuO-ZnO- Al2O3/[gamma]-Al2O3 Catalyst in the Synthesis of	J Erena, I Sierra, M Olazar, AG Gayubo, AT Ind. Eng. Chem, 2008	desorption, temperature is increased from 150 to 550 °C, at a rate of 5 °C min -1 . Copper surface areas (S Cu) for the fresh and deactivated catalysts have been determined by N 2 O chemisorptions in a Micromeritics AutoChem II 2910 coupled to the mass spectrometer	2910
Deactivation of a Pd/Al2O3 catalyst used in hydrodechlorination reactions: Influence of the nature of	Lopez, E., Ordonez, S., Diez, F.V., Applied Catalysis B, Environmental, 62 (1), p.57-65, Jan 2006	measurements were performed in a Micromeritics ASAP 2000 apparatus. Crystallographicstudies were carried out in a Micromeritics TPD-2900 apparatus, equipped with TCDCO in the outlet gas with a mass spectrometer. The results for the CO2signal	2900
Deactivation of Au/CeOx water gas shift catalysts	CH Kim, LT Thompson - Journal of Catalysis, 2005	Temperature-programmed oxidation (TPO) and CO chemisorption were performed with a <u>Micromeritics AutoChem 2910</u> equipped with a Thermo Onix <u>mass spectrometer</u> . The samples were degassed in He at 200 °C prior to analysis	2910
Deactivation of real three way catalysts by CePO4 formationuam.es	C Larese, FC Galisteo, ML Granados, R Applied Catalysis B, , 2003	monochromator. TPR profiles were recorded in a Micromeritics TPD/TPR 2900 equipment HORIBA VIA-510), O 2 by paramagnetic detector (mod. HORIBA MPA-510) and C 3 H 6, H 2 and H 2 O by a quadrupole mass spectrometer (mod. BALTZER	2900
Decarboxylation of fatty acids over Pd supported on mesoporous carbon	I Simakova, O Simakova, P Mäki- Arvela, DY Murzin - Catalysis Today, 2009	CO pulse chemisorption was performed using <u>AutoChem 2910</u> apparatus (<u>Micromeritics</u>). Prior to the measurements the catalysts were reduced similarly as prior to the reaction (see below) The products were identified with a gas chromatograph— <u>mass spectrometer</u> (GC–MS)	2910
Dehydrogenation of long chain paraffins over supported Pt-Sn-K/Al2O3 catalysts: A study of the alumina support effect	He, S., Sun, C., Bai, Z., Dai, X., Wang, B., Applied Catalysis A, General, 356 (1), p.88-98, Mar 2009	adsorption system (Micromeritics ASAP 2010, Americancatalysts (Micromeritics AutoChem II 2920, Americanusing a Micromeritics AutoChem II 2920 apparatusin the Micromeritics AutoChem II 2920 apparatusdetected by a quadruple mass spectrometer (Balzers OmniStar 300	2920
Deoxygenation of palmitic and stearic acid over supported Pd catalysts: Effect of	I Simakova, O Simakova, P Mäki- Arvela, A Applied Catalysis A, , 2009	The product identification was validated with a gas chromatograph— <u>mass spectrometer</u> (GC–MS). 2.3. Catalyst preparation and characterization Metal dispersion was measured also by CO chemisorption using <u>AutoChem</u> <u>2910</u> apparatus (<u>Micromeritics</u>)	2910
Department of Chemical Engineering wvu.edu	CH Clark - 2005 - eidr.wvu.edu	Page 18. 10 require either a TPR apparatus attached to a <u>mass spectrometer</u> or a way of modeling the underlying peak structure (2,3). The characterization methods discussed thus far can be used to characterize the bulk of the material	

Department of Chemical Engineeringwvu.edu	CH Clark - 2005 - eidr.wvu.edu	Page 18. 10 require either a TPR apparatus attached to a <i>mass spectrometer</i> or a way of modeling the underlying peak structure (2,3). The characterization methods discussed thus far can be used to characterize the bulk of the material	
Development of an industrial characterisation method for naphtha reforming	MP Gonzalez- Marcos, B Inarra, JM Guil, MA Gutierrez- Catalysis Today, 2005	Acidity of the catalysts was evaluated by TPD of ammonia and was carried out in a <u>Micromeritics</u> <u>AutoChem 2910</u> instrument The detectors were: an AED, used for quantification, and a <u>mass</u> <u>spectrometer</u> , for identification of the compounds	2910
DEVELOPMENT OF ON- LINE GC/MS MONITORING TECHNIQUES FOR HIGH PRESSURE FUEL CONVERSION PROCESSES	6-Mar	directly coupled to a modified Ion Trap <u>Mass spectrometer</u> (ITMS,Finnigan MAT) with tandem15 C h i n Miniaturized Ion Trap <u>Mass spectrometer</u> (Finnigan-MAT) with a PC comoutercap. 40 C isothermal quadrupole <u>mass spectrometer</u> - modcl HP 5971 (Hewletl- Packard	
doi:10.1016/S0926- 3373(02)00161-3	5-Aug	monochromator. TPR profiles were recorded in a Micromeritics TPD/TPR 2900 equipment. The 100 mg of powder sieved510) and C3H6, H2 and H2O by a quadrupole <u>mass spectrometer</u> (mod. BALTZER Prisma QMS 200 controlled by BALTZER QuadstarTM	2900
Dry reforming of methane over LaNi1− yByO3±δ (B= Mg, Co) perovskites used as	GS Gallego, C Batiot-Dupeyrat, J Barrault, E Applied Catalysis A, , 2007	TPR experiments were carried out in a Micromeritics AutoChem 2910 equipment using about 160 mg of The nitrogen gas adsorption were performed on a Micromeritics ASAP 2000 apparatus at -196 °C The reaction products were analyzed by an on-line mass spectrometer	2910
Dry reforming of methane over LaNi1-yByO3+/-δ (B=Mg, Co) perovskites used as catalyst precursor	Gallego, G.S., Batiot-Dupeyrat, C., Barrault, J., Florez, E., Mondragon, F., Applied Catalysis A, General, 334 (1), p.251-258, Jan 2008	experiments were carried out in a <u>Micromeritics</u> <u>AutoChem 2910</u> equipment using aboutadsorption were performed on a <u>Micromeritics</u> ASAP 2000 apparatus at -196were analyzed by an on-line <u>mass spectrometer</u> . 3 Results and discussion	2910
Dry reforming of methane over nickel catalysts supported on the cuspidine-like	V García, MT Caldes, O Joubert, GS Gallego, C Batiot Catalysis Today, 2008	The temperature-programmed reduction (TPR) studies were performed in a chemisorption unit Micromeritics AutoChem 2910 with samples of 50 mg The composition of the reactants/products mixture was analysed with an on-line mass spectrometer	2910
Dynamic oxygen storage and release over Cu0.1Ce0.9Ox and Cu0.1Ce0.6Zr0.3Ox complex compounds and	Jia, L., Shen, M., Wang, J., Gu, W., Journal of Alloys and Compounds, 473 (1), p.293-297, Apr 2009	monitored with on-line process by quadrupole <u>mass spectrometer</u> (Balzers, QMS200). In this experiment, isothermalreduction (TPR) experiments were performed in an <u>AutoChem 2910 Micromeritics</u> instrument, equipped with a thermal conductivity	2910
Dynamics of N2O decomposition over HZSM-5 with low Fe contentepfl.ch	DA Bulushev, L Kiwi- Minsker, A Renken - Journal of Catalysis, 2004	Catalytic activity measurements. Catalytic activity was measured in a <u>Micromeritics AutoChem 2910</u> analyzer. A ThermoStar 200 (Pfeiffer Vacuum) quadrupole <u>mass spectrometer</u> was used to analyze the gas phase composition	2910
Effect of activated carbon on the dispersion of Ru and K over supported Ru-based catalyst for ammonia synthesis	Han, W., Liu, H., Zhu, H., Catalysis Communications, 8 (3), p.351-354, Mar 2007	N2physisorption was carried on <u>Micromeritics</u> ASAP2010. TPD-MSwas performed on <u>Micromeritics AutoChem 2910</u> attaching withconductivity detector and <u>mass spectrometer</u> detector. Ru andon <u>Micromeritics AutoChem</u> <u>2910</u> . The active	2910

Effect of carbon addition on the Pt-Sn/±c-Al2O3 catalyst for long chain paraffin dehydrogenation to olefin	He, S., Sun, C., Du, H., Dai, X., Wang, B., Chemical Engineering Journal, 141 (1), p.284-289, Jul 2008	adsorption system (<u>Micromeritics</u> ASAP 2010, Americanmercury porosimeter (<u>Micromeritics</u> Autopore 9520, Americancatalysts (<u>Micromeritics</u> <u>AutoChem II 2920</u> , Americancatalysts (<u>Micromeritics</u> <u>AutoChem II 2920</u> , Americandetected by a quadruple <u>mass</u>	2920
Effect of carbon addition on the Pt-Sn/γ-Al2O3 catalyst for long chain paraffin dicp.ac.cn	S He, C Sun, H Du, X Dai, B Wang - Chemical Engineering Journal, 2008	spectrometer. 2.3 Dehydrogenation desorption of ammonia (NH 3 -TPD) experiments were carried out to analyze the acidic properties of the catalysts (Micromeritics AutoChem II 2920 out with a temperature ramp of 10 °C min -1 , and the desorbed products were detected by a quadruple mass spectrometer. 2.3	2920
Effect of carbon nanofiber functionalization on the adsorption properties of volatile organic compounds	Cuervo, M.R., Asedegbega-Nieto, E., Diaz, E., Vega, A., Ordonez, S., Castillejos-Lopez, E., Rodriguez- Ramos, I., Journal of Chromatography A, 1188 (2), p.264- 273, Apr 2008	at -196^oC with a <u>Micromeritics</u> ASAP 2000 surfaceTPO), employing a <u>Micromeritics</u> TPD- <u>2900</u> apparatus connectedPfeiffer Vacuum-300 <u>mass spectrometer</u> (Nashua, NH, USAwith a quadrupole <u>mass spectrometer</u> (Balzers QMG 421-C	2900
EFFECT OF CHLORINE PRECURSOR IN SURFACE AND CATALYTIC PROPERTIES OF Fe/TiO2	9-Jun	recorded with an automatic <u>Micromeritics</u> system ASAP 2001, using nitrogencarried out in a TPR/TPD <u>2900 Micromeritics</u> system provided with a thermalsome experiments a quadrupole <u>mass spectrometer</u> Hiden HPT 20 was used to detect	2900
Effect of Co Content Upon the Bulk Structure of Sr-and Co-doped LaFeO 3	JN Kuhn, US Ozkan - Catalysis Letters, 2008	area measurements were made by the physical adsorption of Kr at 77 K on a Micromeritics ASAP 2010 were conducted on both a Thermo-Finnigan Trace Ultra differential scanning quadrupole (DSQ) gas chromatograph/mass spectrometer (GC/MS) and Auto- chem II 2920	2920
Effect of copper loading on copper-ceria catalysts performance in CO selective oxidation for fuel cell applications	Ayastuy, J.L., Gurbani, A., Gonzalez-Marcos, M.P., Gutierrez- Ortiz, M.A., International Journal of Hydrogen Energy, 35 (3), p.1232-1244, Feb 2010	desorption isotherms at 78K (<u>Micromeritics</u> ASAP 2010). UltravioletCO-TPR was followed by a <u>mass spectrometer</u> (MS) (MKS Cirrus 300experimental equipment (<u>Micromeritics</u> <u>AutoChem</u> <u>2910</u>) with ca. 0.4g of	2910
Effect of Dimethyl Ether Co- feed on Catalytic Performance of Methane dicp.ac.cn	H Chen, Y Li, W Shen, Y Xu, X Bao - Journal of Natural Gas, 2004 - fruit.dicp.ac.cn	TGA profile was recorded automatically. TPO was conducted on a <u>Micromeritics AutoChem II2920</u> instrument equipped with a thermal conductivity detector (TCD) and a <u>mass spectrometer</u> (MS). 0.03 g coked catalyst packed	2920
Effect of H2O and SO2 on the activity of Pd/TiO2 catalysts in catalytic reduction of NO with methane in the presence	Mitome, J., Karakas, G., Bryan, K.A., Ozkan, U.S., Catalysis Today, 42 (1), p.3-11, Jun 1998	the catalysts used were measured by a Micromeritics 21()() E Accusorb instrument using nitrogenoxygen concentrations between 0 and 2900 ppm at 500C. At 2 concentration of 3800CO and between N2O and CO2 using the mass spectrometer. The reaction was first brought to steady	2900
Effect of hydrothermal treatment on the composition and structure of Pt (IV) hydroxo	OB Bel'skaya, VA Drozdov, TI Gulyaeva, AB Kinetics and, 2009	and thermal decomposition products were analyzed on an STA 449 C Jupiter ther mal analyzer (Netzsch) connected to a QMS 403 C Aeolos quadrupole <u>mass spectrometer</u> with a Before measurements with an <u>AutoChem</u> 2920 instrument (<u>Micromeritics</u>), the sample was	2920

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Effect of La2O3 doping on syntheses of C1–C18 mixed linear α -alcohols from	G Jiao, Y Ding, H Zhu, X Li, J Li, R Lin, W Dong, Applied Catalysis A, , 2009	TPR experiments were performed on a <u>Micromeritics AutoChem 2910</u> apparatus The hydrogen consumption was recorded with a thermal conductivity detector (TCD) while some effluent was simultaneously traced with an Omnistar 300 quadrupole <u>mass spectrometer</u> for analysis	2910
Effect of MgO addition on the basicity of Ni/ZrO2 and on its catalytic activity in carbon dioxide reforming of methane	V García, JJ Fernández, W Ruíz, F Mondragón, A Catalysis, 2009	NiO x /ZrO 2 –MgO) was studied by temperature-programmed reduction (TPR) in a chemisorption unit (<u>AutoChem 2910</u> , <u>Micromeritics</u>) mL min –1). Carbon dioxide desorbed from the sample was online monitored with a quadrupole-type <u>mass spectrometer</u> (Thermo Onix	2910
Effect of Nitric Acid Treatment on Carbon Nanotubes (CNTs)- Cordierite Monoliths	X Yu, B Lin, B Gong, J Lin, R Wang, K Wei - Catalysis Letters, 2008	CO chemisorption was carried out with an <u>AutoChem 2910</u> instrument (<u>Micromeritics</u>) The effluents (m/e = 28 for CO, 44 for CO 2) were monitored by an online <u>mass spectrometer</u> (Pfeiffer vacuum, OmniStar). 2.4 Activity Studies	2910
Effect of Organic Nickel Precursor on the Reduction Performance and	F Li, X Yi, W Fang - Catalysis Letters, 2009	The temperature-programmed desorption (TPD) exper- iments were performed on a Micromeritics AutoChem II 2920 instrument C min -1 in a flow of Ar (40 mL min -1). H 2 desorption was monitored by an online ThermoStar quadrupole <u>mass</u> <u>spectrometer</u> (model GSD301T2)	2920
Effect of Pt Impregnation on a Precipitated Iron- based Fischer–Tropsch Synthesis	W Yu, B Wu, J Xu, Z Tao, H Xiang, Y Li - Catalysis Letters, 2008	CO temperature-programmed reduction (CO-TPR) experiment was carried out in a Micromeritics AutoChem G 2920 analyzer and an on-line mass spectrometer (QIC20). In CO-TPR experiment, about 100 mg of catalyst was loaded in a U-shape quartz tube flow reactor	2920
Effect of Rh loading on the performance of Rh/Al2O3 for methane partial oxidation to synthesis gas	Li, J.M., Huang, F.Y., Weng, W.Z., Pei, X.Q., Luo, C.R., Lin, H.Q., Huang, C.J., Wan, H.L., Catalysis Today, 131 (1), p.179-187, Feb 2008	quadrupole <u>mass spectrometer</u> (PfeifferoC using a <u>Micromeritics</u> Tristar 3000performed by a <u>Micromeritics AutoChem II 2920</u> instrumentquadrupole <u>mass spectrometer</u> (Pfeifferquadrupole <u>mass spectrometer</u> for the reaction	2920
Effect of Ru on LaCoO3 perovskite-derived catalyst properties tested in oxidative reforming of diesel	Navarro, R.M., Alvarez-Galvan, M.C., Villoria, J.A., Gonzalez-Jimenez, I.D., Rosa, F., Fierro, J.L.G., Applied Catalysis B, Environmental, 73 (3), p.247-258, May 2007	measurements were performed with a Micromeritics ASAP 2100 apparatus on samplescatalysts were conducted using a Micromeritics 2900 instrument in a U-shaped quartzon-line with a quadrupole mass spectrometer (Balzers QMS 200), allowing	2900
Effect of structural and acidity/basicity changes of CuO–CeO2 catalysts on their	P Djinović, J Levec, A Pintar - Catalysis Today, 2008	of a U-shaped quartz test tube (designed to minimize channeling effects), which was inserted into an electric furnace of the <u>Micromeritics'</u> <u>AutoChem</u> apparatus WGS reaction products as well as unconverted reactants were continuously recorded with a <u>mass spectrometer</u>	AutoCh em
Effect of the nature of the support on the enantioselective hydrogenation of [86K]	TERESITA MARZIALETTI, J.L.G. FIERRO, P. REYES, Facultad de Ciencias, Jun 2009	was studied in a TPR/TPD <u>2900 Micromeritics</u> system equipped with a thermal2 at 77 K in an automatic <u>Micromeritics</u> system Model ASAP 2010. Theanalyzed in a gas chromatograph- <u>mass spectrometer</u> (GCMS-QP5050 Shimadzu	2900

Effect of the preparation technique on the catalytic performances of TiO2 supported	MP Casaletto, L Lisi, G Mattogno, P Patrono, F Pinzari, Catalysis Today, 2004	Temperature programmed reduction (TPR) with H 2 and temperature programmed desorption (TPD) of NH 3 were carried out using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyser equipped with a TC detector and coupled with a Hiden HPR 20 <u>mass spectrometer</u> as described in [3	2900
Effect of the reductant nature on the catalytic removal of N2O on Fezeolite-β	G Delahay, M Mauvezin, A Guzmán-Vargas, B Catalysis, 2002	Temperature programmed reduction analysis (TPR) by H 2 /Ar (3/97) was carried out with a Micromeritics AutoChem 2910 apparatus using TCD For these TPR experiments, the detection was processed by a Pfeiffer Omnistar QMS 200 mass spectrometer and the masses 2 (H 2	2910
Effect of the Reduction Temperature of Co-La- Zr/AC on the Synthesis of Higher	G JIAO, Y DING, H ZHU, X LI, W DONG, J LI, Y Chinese Journal of , 2009	XRD analysis was carried out using a PANalytical X Pert PRO diffractometer. Temperature-programmed reduction (TPR) experiments were performed on a <u>Micromeritics AutoChem 2910</u> apparatus, and an Omnistar 300 quadrupole <u>mass spectrometer</u> was used for detection	2910
Effect of the support on the kinetic and deactivation performance of Pt/support catalysts during coupled hydrogenation	Castano, P., Gutierrez, A., Pawelec, B., Fierro, J.L.G., Aguayo, A.T., Arandes, J.M., Applied Catalysis A, General, 333 (2), p.161-171, Dec 2007	obtained in a <u>Micromeritics</u> ASAP 2010 instrument <u>Micromeritics</u> TPO/TPR <u>2900</u> apparatus. Beforerecorded using a <u>mass spectrometer</u> quadrupoleattached to a <u>mass spectrometer</u> quadrupolerecorded by the <u>mass spectrometer</u> and the area	2900
Effect of ultrasonic power on the structure of activated carbon and the activities of	F Yu, J Ji, Z Xu, H Liu - Ultrasonics, 2006	Surface oxygen groups analysis was performed employing a TPD–MS equipment (<u>Micromeritics AutoChem 2910</u>) with a quartz reactor containing 100 mg catalyst 1 . Desorption products of CO (m/e = 28) and CO 2 (m/e = 44) were detected by an Omnistar <u>mass spectrometer</u>	2910
Effect of ultrasound in enantioselective hydrogenation of 1-phenyl-1, 2	B Toukoniitty, E Toukoniitty, P Mäki- Arvela, JP Ultrasonics, 2006	Temperature programmed desorption (TPD) of hydrogen was carried out with an AutoChem 2910 instrument (Micromeritics). The desorbed gases were identified and analyzed by a TC detector and a quadropole <u>mass spectrometer</u> (Carlo Erba Instruments)	2910
Effect of vanadium on the behaviour of unsulfated and sulfated Ti-pillared clay	J Arfaoui, LK Boudali, A Ghorbel, G Delahay - Catalysis Today, 2009	by temperature-programmed desorption (TPD) of ammonia using an <u>AUTOCHEM 2910</u> (<u>Micromeritics</u>) programmed reduction have been carried out in the <u>AUTOCHEM 2910</u> with a were analysed by sampling on line with a quadruple <u>mass spectrometer</u> (Pfeiffer Omnistar	2910
Effects of calcination temperatures on the catalytic performance of Rh/Al2O3 for methane partial oxidation to	WZ Weng, XQ Pei, JM Li, CR Luo, Y Liu, HQ Lin, CJ Catalysis Today, 2006	The TPSR experiments were performed by a <u>Micromeritics AutoChem II 2920</u> instrument using CH 4 /O 2 /He = 2/1/45 mixture (in the volume ratio) as The products of the TPSR reaction were analyzed by an on-line ThermoStar quadrupole <u>mass spectrometer</u> (GSD301T2)	2920
Effects of CeO2 addition on Ni/Al2O3 catalysts for the reaction of ammonia	W Zheng, J Zhang, Q Ge, H Xu, W Li - Applied Catalysis B, Environmental, 2008	A <u>Micromeritics</u> ASAP 2010P automated physisorption instrument was used to measure the N 2 NH 3 -TPSR experiments were carried out in a commercial Micrometric <u>AutoChem</u> <u>2910</u> apparatus K. The TPSR process was monitored by an on-line <u>mass spectrometer</u> (Omnisorp	2910

Effects of palladium loading on the response of a thick film flame-made ZnO mdpi.org	C Liewhiran, S Phanichphant - Sensors, 2007 - mdpi.org	Page 7. Sensors 2007, 7 1165 He, <u>Micromeritics</u> <u>AutoChem II 2920</u> unit) reached. The recovery times, T rec denotes the time needed until 90% of the original baseline signal is recovered [52, 53]. Furthermore, the experimental set up had a <u>mass spectrometer</u> (MS) connected for	2920
Enhanced catalytic activity for butane isomerization with alumina-promoted tungstated mesoporous zirconia	Hwang, C.C., Chen, X.R., Wong, S.T., Chen, C.L., Mou, C.Y., Applied Catalysis A, General, 323, p.9- 17, Apr 2007	porosity data were determined at 77K on a Micromeritics ASAP 2010 instrument. Ultraviolet- visiblesamples was carried out on Micrometrics AutoChem 2910 instrument. Before NH3adsorptionprocess was monitored by a Quadruple Mass spectrometer (Thermo ONIX ProLab) using the mass	2910
Enhanced Hydrogen Storage Performance of LiBH4-SiO2-TiF3 Compositedicp.ac.cn	Y Zhang, WS Zhang, MQ Fan, SS Liu, Journal of Physical , 2008 - taozhang.dicp.ac.cn	TPD was conducted on Micromeritics AutoChem II under 0.1 MPa argon, to which a mass spectrometer (Omnistar) was attached for characterizing the liberated gases. TG measurements were executed upon TherMax 500 (Thermo Cahn)	2920
Enhancement of the CO2 retention capacity of X zeolites by Na- and Cs-treatments	Diaz, E., Munoz, E., Vega, A., Ordonez, S., Chemosphere, 70 (8), p.1375-1382, Feb 2008	at -196^oC on a <u>Micromeritics</u> ASAP 2000 instrumentby NH3-TPD, in a <u>Micromeritics</u> TPD- <u>2900</u> apparatus connected to a Glaslab 300 <u>mass spectrometer</u> using He as thebaseline of the <u>mass spectrometer</u> . The TPD tests were	2900
Ethanol steam reforming and water gas shift reaction over Co–Mn/ZnO catalysts	A Casanovas, C de Leitenburg, A Trovarelli, J Chemical Engineering, 2009	Temperature programmed reduction (TPR) was carried out with a <u>Micromeritics AutoChem II 2920</u> instrument using a H 2 /Ar mixture (5% H 2) at 10 K min –1 and a TCD detector The effluent of the reactor was monitored on line with a MKS Cirrus <u>mass spectrometer</u>	2920
Ethanol steam reforming over Co-based catalysts: Role of oxygen mobility	H Song, US Ozkan - Journal of Catalysis, 2009	Temperature-programmed oxidation (TPO) experiments were performed using <u>AutoChem-2920</u> (<u>Micromeritics</u>) with an online <u>mass spectrometer</u> (MS) (MKS Instruments, 1–300 amu). The samples were first pretreated at 300	2920
Evaluating the Catalytic Performances of SAPO-34 Catalysts for the Oxidative	L Lisi, L Marchese, HO Pastore, A Frache, G Topics in Catalysis, 2003	Temperature-programmed desorption of NH3 (NH3-TPD) measurements were carried out on the activated catalysts using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyzer equipped with a TC detector and coupled with a Hiden HPR 20 <u>mass spectrometer</u>	2900
Evaluation of different zeolites in their parent and protonated forms for the catalytic combustion of hexane and benzene	Diaz, E., Ordonez, S., Vega, A., Coca, J., Microporous and Mesoporous Materials, 83 (1), p.292-300, Sep 2005	adsorption at -196^oC with a <u>Micromeritics</u> ASAP 2000 surface analyserstudies were carried out using a <u>Micromeritics</u> TPD- <u>2900</u> apparatus connected to a massusing a Glaslab 300 quadrupole <u>mass spectrometer</u> , which used a capillary inlet	2900
Evolution of the properties of PtGe/Al2O3 reforming catalysts with Ge content	R Mariscal, JLG Fierro, JC Yori, JM Parera, JM Applied Catalysis A, , 2007	Temperature-programmed reduction (TPR) profiles were taken on a semiautomatic <u>Micromeritics</u> TPD/TPR <u>2900</u> apparatus interfaced with a microcomputer heating rate of 10 °C min -1. The desorbing species were detected with a quadrupole <u>mass spectrometer</u> (Balzers QMG	2900
Experimental techniques for investigating the surface oxygen formation in the N2O	A Ates, A Reitzmann - Chemical Engineering Journal, 2007	studies of N 2 O decomposition were performed in a set-up for catalyst characterisation (<u>AutoChem 2910</u> , <u>Micromeritics</u>) containing a quartz glass reactor (id 9 mm). The reaction products were monitored in an on-line mode using a quadrupole <u>mass spectrometer</u> (QMS 422	2910
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Factors affecting isomer yield for n-heptane hydroisomerization over as	S Gopal, PG Smirniotis - Journal of Catalysis, 2004	The analysis was carried out using a Micromeritics AutoChem 2910 automated catalyst characterization system Product identification was accomplished using a gas chromatograph (Hewlett-Packard, 5890 Series II) equipped with a mass spectrometer (Hewlett-Packard, 5972	2910
Fe-Ce-ZSM-5 a new catalyst of outstanding properties in the selectiversc.org	G Carja, G Delahay, C Signorile, B Coq - Chemical Communications, 2004	Temperature programmed reduction (TPR) by H2/Ar (3/97) was carried out by using a Micromeritics AutoChem 2910 apparatus The effluent composition was monitored continuously by sampling on line to a quadrupole mass spectrometer (Pfeiffer vacuum Omnistar equipped	2910
Fine tuning the surface acid/base properties of single step flame-made Pt/alumina	B Schimmoeller, F Hoxha, T Mallat, F Krumeich, Applied Catalysis A: , 2009	Pt metal dispersion was determined by CO-pulse chemisorption on a <u>Micromeritics AutoChem II</u> <u>2920</u> unit. Off-gas was analyzed via a <u>mass</u> <u>spectrometer</u> (Pfeiffer Vacuum, Thermostar) to derive the amount of chemisorbed CO [27]	2920
Fischer-Tropsch synthesis over ±c-alumina-supported cobalt catalysts: Effect of support variables	Borg, O., Eri, S., Blekkan, E.A., Storsaeter, S., Wigum, H., Rytter, E., Holmen, A., Journal of Catalysis, 248 (1), p.89-100, May 2007	Nitrogen adsorption/desorption isotherms were measured on a <u>Micromeritics</u> TriStar 3000 instrument, and the data were collected atchemisorption Hydrogen adsorption isotherms were recorded on a <u>Micromeritics</u> ASAP 2010 unit at 312 K. The samples (0.5 g, 53-90 mum	
Fischer–Tropsch synthesis over γ-alumina-supported cobalt catalysts: Effect of	Ø Borg, S Eri, EA Blekkan, S Storsæter, H Wigum, E Journal of Catalysis, 2007	For one catalyst (C-11), TPR was performed on a Micromeritics AutoChem II instrument connected to a Pfeiffer Vacuum ThermoStar mass spectrometer. The sample was subjected to 10% H 2 in Ar while the temperature was increased from 300 to 1353 K at 10 K/min	2920
Flame-derived Pt/Ba/CexZr1- xO2: Influence of support on thermal deterioration	R Strobel, F Krumeich, SE Pratsinis, A Baiker - Journal of Catalysis, 2006	programmed decomposition (TPD) of BaCO 3 was measured using a <u>Micromeritics AutoChem II</u> 2920 by heating (10 °C/min) 35 mg of powder in a helium flow (20 ml/min) from 50 to 1000 °C. Gasphase composition was monitored by a <u>mass spectrometer</u> (Thermostar, Pfeiffer	2920
Flame-derived Pt/Ba/CexZr1-xO2: Influence of support on thermal deterioration and behavior as NOx	Strobel, R., Krumeich, F., Pratsinis, S.E., Baiker, A., Journal of Catalysis, 243 (2), p.229-238, Oct 2006	the BET method (Micromeritics Tristar). Beforemeasured using a Micromeritics AutoChem II 2920 by heatingmonitored by a mass spectrometer (Thermostar, Pfeifferanalyzed using a mass spectrometer (Thermostar, Pfeiffer	2920
Flame-made Pt-Ba/Al2O3 catalysts: Structural properties and behavior in lean-NOx storage-reduction	Piacentini, M., Strobel, R., Maciejewski, M., Pratsinis, S.E., Baiker, A., Journal of Catalysis, 243 (1), p.43-56, Oct 2006	77 K using a <u>Micromeritics</u> Tristar 3000 <u>Micromeritics AutoChem II 2920</u> instrument. A <u>mass spectrometer</u> (Pfeiffer VacuumVacuum GSD 301 O1 <u>mass spectrometer</u> , which was connectedcalibration of the <u>mass spectrometer</u> (injection of	2920

Flame-made Pt–Ba/Al2O3 catalysts: Structural properties and behavior in	M Piacentini, R Strobel, M Maciejewski, SE	CO-pulse chemisorption was performed by	
lean	Pratsinis, Journal of Catalysis, 2006	injecting pulses (0.35 ml) of 10% CO/He into 5% H 2 /Ar at 40 °C on a Micromeritics AutoChem II 2920 instrument. A <i>mass spectrometer</i> (Pfeiffer Vacuum, Thermostar) was used to analyze the off-gas and derive the amount of	2920
Formation of the Surface NO during N2O Interaction at Low Temperature with Ironepfl.ch	DA Bulushev, A Renken, L Kiwi- Minsker - J. Phys. Chem. B, 2006	Transient Response and TPD Measurements. Transient response and TPD experiments were performed in a Micromeritics AutoChem 2910 analyzer. A ThermoStar 200 (Pfeiffer Vacuum) quadrupole mass spectrometer was used for gas analysis	2910
Ga-promoted tungstated zirconia catalyst for n-butane isomerization	XR Chen, CL Chen, NP Xu, S Han, CY Mou - Catalysis Letters, 2003	The temperature-programmed desorption of ammonia (NH3 TPD) was carried out using a Micromeritics Auto- Chem 2910 instrument The desorption process was monitored by a quadruple mass spectrometer (Thermo ONIX, ProLab) connected on- line through a heated	2910
Gas chromatography / mass spectrometry analysis of components of pyridine temperature-programmed desorption spectra from	Pribylova, L., Dvorak, B., Journal of Chromatography A, 1216 (18), p.4046-4050, May 2009	TPD apparatus <u>AutoChem</u> <u>2920</u> with thevery sensitive <u>mass spectrometer</u> (MS). Applicationcommercial device <u>AutoChem</u> <u>2920</u> from <u>Micromeritics</u> corp. Two peripheraljoined to the <u>AutoChem</u> <u>2920</u> . The gasequipped with a <u>mass spectrometer</u> for the qualitative	2920
Gas-phase dehydration of glycerol over ZSM-5 catalysts	YT Kim, KD Jung, ED Park - Microporous and Mesoporous Materials, 2009	programmed oxidation (TPO) was conducted over 0.05 g of the sample in a 2% O 2 /He stream by heating the sample from 30 to 800 °C at a heating rate of 10 °C/min while monitoring the TCD signal (<u>AutoChem 2910</u> unit, <u>Micromeritics</u>) and on-line <u>mass spectrometer</u> (QMS 200	2910
Gas-phase hydrogenation of o-xylene over Pt/alumina catalyst, activity, and	A Kalantar Neyestanaki, P Mäki-Arvela, H Backman, Journal of Catalysis, 2003	Temperature-programmed desorption (TPD) of hydrogen and o-xylene was carried out with an AutoChem 2910 instrument (Micromeritics). The desorbed gases were identified and analyzed by a TC detector and a quadropole mass spectrometer (Omnistar, Baltzer Instruments)	2910
Gas-phase hydrogenation of o-xylene over Pt/knitted silica-fiber catalysts	AK Neyestanaki, P Maki-Arvela, H Backman, H Ind. Eng. Chem, 2003	programmed desorption (TPD) of ammonia, hydrogen, and o-xylene was carried out in a volumetric equipment (<u>AutoChem 2910</u> , <u>Micromeritics</u>) in the Desorbed gases were identified and analyzed by a quadruple <u>mass spectrometer</u> (Omnistar, Baltzer Instruments)	2910
Highly active structured catalyst made up of mesoporous Co3O4 nanowires	G Marbán, I López, T Valdés-Solís, AB International Journal of, 2008	Nitrogen adsorption isotherms were performed at -196 °C on a <u>Micromeritics</u> ASAP 2020 volumetric adsorption system TPR analyses were performed in a chemisorption analyzer (<u>AutoChem II</u>) equipped with a TCD reactor and a <u>mass</u> <u>spectrometer</u> (OmniStar 300O)	2920
Highly dispersed gold on activated carbon fibers for low-temperature CO oxidation	DA Bulushev, I Yuranov, EI Suvorova, PA Buffat, L Journal of Catalysis, 2004	in He (100 ml/min, ramp rate 10 K/min) using a Micromeritics AutoChem 2910 analyzer. In these experiments 0.010 g of ACF was placed in a quartz plug-flow reactor. The TPD products were analyzed by a ThermoStar-200 quadrupole mass spectrometer (Pfeiffer Vacuum	2910
Highly dispersed sol-gel synthesized Cu-ZrO2 materials as catalysts for oxidative	S Esposito, M Turco, G Bagnasco, C Applied Catalysis A: , 2009	samples treated in air flow at 340 °C using a 2% H 2 /Ar mixture and an heating rate of 10 °C min −1 with a <u>Micromeritics</u> <u>2900</u> apparatus TCD detector allowed the analysis of H 2 , CO (detection limit = 0.01%) CO 2 , O 2 , CH 4 , CH 3 OH, H 2 O. A <u>mass spectrometer</u> Hiden was	2900

Highly dispersed solgel synthesized CuZrO2 materials as catalysts for oxidative steam reforming of methanol	Esposito, S., Turco, M., Bagnasco, G., Cammarano, C., Pernice, P., Aronne, A., Applied Catalysis A, General, 372 (1), p.48-57, Jan 2010	at 77K were obtained by a Micromeritics Gemini II 2370 apparatusrate of 10^oCmin-1with a Micromeritics 2900 apparatus. Copper dispersionO2, CH4, CH3OH, H2O. A mass spectrometer Hiden was employed for identification	2900
Highly efficient heterogenous catalyst for acylation of alcohols and amines using	B Sreedhar, R Arundhathi, MA Reddy, G Applied Clay, 2009	Programmed Desorption (TPD) studies were conducted on an AutoChem 2910 Micromeritics area measurements of samples were performed on a Micromeritics ASAP2020 automated sample were monitored online by a quadrapole <u>mass spectrometer</u>	2910
HOT GAS DESULFURIZATION BY ZINC OXIDE-TITANIUM DIOXIDE REGENERABLE SORBENTS.	6-Feb	RU300 inshument. Specific surface area was measured with a <u>Micromeritics</u> Flow Sorb II 2300 BET apparatus. The pore volume and pore size distribution were detemined by 3 <u>Micromeritics</u> Autopore 9200. Reactions were performed in a Cahn 113-X	
Hydrogen peroxide decomposition over Ln1- xAxMnO3 (Ln= La or Nd and A= K or	YN Lee, RM Lago, JLG Fierro, J González - Applied Catalysis A, General, 2001	Temperature programmed reduction (TPR) and O 2 temperature programmed desorption (TPO) profiles were obtained in a <u>Micromeritics 2900</u> instrument at 10 K min -1 and the desorption products O 2, CO 2 and H 2 O monitored by a <u>mass spectrometer</u> detector Balzers QMG	2900
Hydrogen production by steam reforming of vegetable oils using nickelbasedurv.es	M Marquevich, X Farriol, F Medina, D Montane - Ind. Eng. Chem. Res, 2001	Hydrogen chemisorption was measured by pulses with a Micromeritics AutoChem 2910 instrument equipped with a TCD detector qualitatively to monitor the extent of cracking by GC-MS using a Hewlett-Packard 5890 chromatograph with a 5989A mass spectrometer system	2910
HYDROGEN TRANSFER PROPERTIES OF SOME COAL PROCESS RECYCLE SOLVENTS.	6-Mar	101, and a flame ionization detector. Gc-ms data were obtained using a Hewlett-Packard 59958 gas chromatograph/mass spectrometer f i t t e d with the same column. Exchange experiments were done using three different Gc analyses were obtained 1 deuterium	
Hydroisomerization in liquid phase of a refinery naphtha stream over Pt- Ni/H-beta zeolite catalysts	Funez, A., De Lucas, A., Sanchez, P., Ramos, M.J., Valverde, J.L., Chemical Engineering Journal, 136 (2), p.267-275, Mar 2008	data acquired on a <u>Micromeritics</u> ASAP 2010 apparatusammonia (TPDA) using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyzer. The sample <u>Micromeritics</u> TPD/TPR <u>2900</u> analyzer). AfterSHIMADZU) coupled to a <u>mass spectrometer</u> (QP-5000 SHIMADZU	2900
Hydroisomerization of n- hexane over gallium- promoted sulfated zirconia	C Cao, S Han, CL Chen, NP Xu, CY Mou - Catalysis Communications, 2003	desorption (TPD) of ammonia was carried out on a <u>Micromeritics AutoChem 2910</u> instrument. TPD profile of ammonia was obtained from 120 to 800 °C at a heating rate of 10 °C/min. The desorption process was monitored by a Quadruple <u>Mass spectrometer</u> (Thermo ONIX	2910
Hydrothermal Fabrication and Catalytic Properties of YBa 2 Cu 3 O 7 Single	Y Zhang, L Zhang, J Deng, H Dai, H He - Catalysis Letters	Hydrogen temperature-programmed reduction (H2-TPR) experiments were conducted on a Micromeritics AutoChem II 2920 chemical adsorption apparatus The outlet gases were analyzed on-line by a <i>mass spectrometer</i> (Hiden HPR20)	2920

Identification of Iron Species in Fe- BEA: Influence of the Exchange Level	M Mauvezin, G Delahay, B Coq, S Kieger, JC J. Phys. Chem, 2001	TPR by H 2 was carried out with a <u>Micromeritics</u> <u>AutoChem</u> <u>2910</u> apparatus using TCD detection For these TPR experiments, the detection was processed by a Pfeiffer Omnistar QMS 200 <u>mass spectrometer</u> , and the masses 2 (H 2), 18 (H 2 O), 28 (N 2 or CO), 30 (NO), 32 (O 2	2910
Implication of the acid-base properties of V/Ti-oxide catalyst in toluene partial oxidation	Kiwi-Minsker, L., Bulushev, D.A., Rainone, F., Renken, A., Journal of Molecular Catalysis A: Chemical, 184 (1), p.223-235, Jun 2002	Balzers QMG-421 <u>mass spectrometer</u> and a Perkin-Elmerfactors of the <u>mass spectrometer</u> for carbon oxidesperformed via a <u>Micromeritics</u> <u>AutoChem</u> <u>2910</u> analyser with200 quadrupole <u>mass spectrometer</u> (Pfeiffer Vacuum	2910
Implication of the acid- base properties of V/Ti- oxide catalyst in toluene partialepfl.ch	L Kiwi-Minsker, DA Bulushev, F Rainone, A Journal of Molecular , 2002	Temperature-programmed reduction (TPR) experiments were performed via a <u>Micromeritics</u> <u>AutoChem 2910</u> analyser with a quartz plug-flow reactor. Hydrogen concentration was determined by a Thermostar 200 quadrupole <u>mass spectrometer</u> (Pfeiffer Vacuum)	2910
In situ Raman study on the partial oxidation of methane to synthesis gas over Rh/	Y Liu, FY Huang, JM Li, WZ Weng, CR Luo, ML Journal of Catalysis, 2008	The reaction products were analyzed by an online	2920
		Balzers OmniStar quadrupole <u>mass spectrometer</u> (model QMS 200). 2.3 The O 2 temperature-programmed desorption (O 2 -TPD) experiments were performed with a <u>Micromeritics AutoChem II 2920</u> instrument. The catalyst (ThermoStar quadrupole <u>mass spectrometer</u> (model GSD301T2	2920
Independent control of metal cluster and ceramic particle characteristics during one	H Schulz, L Madler, R Strobel, R Jossen, SE Journal of Materials, 2005 - mrs.org	Micromeritics AutoChem II 2920) by heating the particles in O2 (20 ml min-1, Pan Gas, 99.999%) up to 900 °C with 10 °C min-1 and moni- toring the evolving CO2 and CO with a mass spectrom- eter (ThermoStar, Asslar, Germany, Pfeiffer Vacuum). The mass spectrometer	2920
INFLUENCE OF BASIC PROPERTIES OF Mg, AI- MIXED OXIDES ON THEIR sbq.org.br	CN Pérez, JLF Monteiro, JML Nieto, CA Henriques - quimicanova.sbq.org .br	TPD analyses were run under He at a heating rate of 20 K min-1 up to 723 K. The amount of CO2 chemisorbed and its desorption profile were measured on a <u>Micromeritics 2900 TPR/TPD</u> analyzer with a quadrupole <u>mass spectrometer</u> detector (Balzers QMS-200)	2900
Influence of catalyst treatments on the adsorption properties of ±c- Al2O3 supported Pt, Rh and Ru catalysts	Diaz, E., Ordonez, S., Vega, A., Coca, J., Microporous and Mesoporous Materials, 77 (2), p.245-255, Jan 2005	nitrogen adsorption with a <u>Micromeritics</u> ASAP 2000 apparatus) andexperiments were carried out in a <u>Micromeritics</u> TPD/TPR <u>2900</u> apparatus connected to anreleased HCl, CO 2 and NO with a <u>mass spectrometer</u> , as the main metal precursors	2900
Influence of clay binder on the liquid phase hydroisomerization of n- octane over	A De Lucas, P Sánchez, A Fúnez, MJ Ramos, Journal of Molecular , 2006	of the acid sites was measured by temperature programmed desorption of ammonia (TPDA) using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyzer Liquid products were analyzed in a gas chromatograph (GC-17A SHIMADZU) coupled to a <u>mass</u> <u>spectrometer</u> (QP-5000 SHIMADZU)	2900

Influence of iron promoter on catalytic properties of Rh-Mn-Li/SiO2 for CO hydrogenation	Yin, H., Ding, Y., Luo, H., Zhu, H., He, D., Xiong, J., Lin, L., Applied Catalysis A: General, 243 (1), p.155-164, Mar 2003	was performed on an America <u>Micromeritics</u> <u>AutoChem 2910</u> . Eighty milligrams of2 O that was provided by <u>Micromeritics</u> . 2.3.2 CO uptakes The apparatusml/min) with a quadrupole <u>mass spectrometer</u> (QMS, Balzers OmniStar 300	2910
Influence of lanthanum on the performance of Zr- Co/activated carbon catalysts in	T Wang, Y Ding, Y Lü, H Zhu, L Lin - Journal of Natural Gas Chemistry, 2008	Temperature programmed desorption of adsorbed CO Temperature programmed desorption of adsorbed CO (CO-TPD) was carried out on a flow apparatus of America Micromeritics AutoChem 2910 The product was detected simultaneously by a quadrupole mass-spectrometer	2910
Influence of Morphological, Redox and Surface Acidity Properties on WGS Activity 	P Djinovic, J Batista, J Levec, A Pintar - JOURNAL OF CHEMICAL, 2009 - J-STAGE	Decomposition of N2O and formation of N2 were monitored with a <u>mass spectrometer</u> from Pfeiffer-Vacuum (model ThermoStar) quartz test tube (designed to minimize channeling ef- fects), which was inserted into an electric furnace of the <u>Micromeritics AutoChem II</u> 2920	2920
Influence of oxidation on heat-treated activated carbon support properties and	H Zhu, W Han, H Liu - Catalysis Letters, 2007	The ash content was determined by heating in air at 800 °C to constant weight. The surface functional groups were determined by TPD-MS, which was performed on a <u>Micromeritics AutoChem</u> <u>2910</u> attached to a QMS 200 (Omnistar) <u>mass</u> <u>spectrometer</u>	2910
Influence of Potassium Doping on the Formation of Vanadia Species in V/Ti Oxideepfl.ch	DA Bulushev, F Rainone, L Kiwi- Minsker, A Renken - Langmuir, 2001	Temperature-Programmed Reduction. A Micromeritics AutoChem 2910 analyzer with a quartz plug-flow reactor was used for the TPR studies. The products in the reactor outlet were analyzed by a ThermoStar quadrupole mass spectrometer (Pfeiffer Vacuum)	2910
Influence of Pr and Ce in dry methane reforming catalysts produced from La1-xAxNiO3-δ perovskites	Gallego, G.S., Marin, J.G., Batiot- Dupeyrat, C., Barrault, J., Mondragon, F., Applied Catalysis A, General, 369 (1), p.97-103, Nov 2009	experiments were carried out in a Micromeritics AutoChem 2910 using about 160mg ofisotherms were obtained in a Micromeritics Flowsorb II 2300 apparatusanalyzed by an on-line mass spectrometer. The reproducibility of	2910
Influence of preparation methods of LaCoO3 on the catalytic performances in the	JP Dacquin, C Lancelot, C Dujardin, P Da Applied Catalysis B,, 2009	H 2 -temperature-programmed reduction experiments (H 2 -TPR) were carried out in a Micromeritics AutoChem II 2920 with 5 vol.% H 2 in Ar and a gradual heating rate of 5 °C The outlet gas mixture was simultaneously analyzed using a GEV 010 Omnistar <u>mass spectrometer</u>	2920
Influence of Pt location on BaCO3 or Al2O3 during NOx storage reduction	R Büchel, R Strobel, F Krumeich, A Baiker, SE Journal of Catalysis, 2009	The Pt dispersion was measured by CO-pulse chemisorption at 40 °C on a <u>Micromeritics</u> <u>AutoChem II 2920</u> to 40 °C. Pulses of 0.35 mL 10% CO/He were injected in 10% H 2 /Ar and the CO concentration in the off gas was recorded using a <u>mass spectrometer</u> (Pfeiffer Vacuum	2920
Influence of textural properties of activated carbons on Pd/carbon catalysts synthesis for cinnamaldehyde hydrogenation	Cabiac, A., Cacciaguerra, T., Trens, P., Durand, R., Delahay, G., Medevielle, A., Plee, D., Coq, B., Applied Catalysis A, General, 340 (2), p.229-235, Jun 2008	77K using a Micromeritics ASAP 2000quadruple mass spectrometer (Pfeifferapparatus (AutoChem 2910, Micromeritics), eventuallyquadrupole mass spectrometer (Pfeiffer2010 Chemi, Micromeritics) and using	2910

Influence of the AI source and synthesis of ordered AI-SBA-15 hexagonal particles	W Li, SJ Huang, SB Liu, MO Coppens - Langmuir, 2005	The acidic properties of the products were examined using temperature programmed desorption (TPD) of ammonia (Micromeritics TPD/TPR 2900) flow for 25 min to remove physically adsorbed NH 3 . The NH 3 TPD spectra were detected by a mass spectrometer by increasing	2900
Influence of the nature of titanium source and of vanadia content on the properties of titanium- pillared montmorillonite	Arfaoui, J., Boudali, L.K., Ghorbel, A., Delahay, G., Journal of Physics and Chemistry of Solids, 69 (5), p.1121-1124, May 2008	carried out at 77K using a Micromeritics ASAP 2000 and ATG, on aAr were carried out in an AUTOCHEM 2910 (Micromeritics). Before catalytic testsO2]=3%) was monitored by mass spectrometer between 50 and 450^oC. 3	2910
Influence of the preparation method on the properties of Fe-ZSM-5 for the selective catalytic reduction of NO by	Guzman-Vargas, A., Delahay, G., Coq, B., Lima, E., Bosch, P., Jumas, J.C., Catalysis Today, 107, p.94-99, Oct 2005	N2physisorption at 77K on a Micromeritics ASAP 2100 instrument, X-rayH2was carried out using a Micromeritics AutoChem 2910 instrument with thermalsampling linked to a quadruple mass spectrometer (Pfeiffer Omnistar) equipped	2910
Influence of thermal treatments on the basic and catalytic properties of Mg, scielo.br	RBIIV Zonno, IAV Santos, CA Henriques, JLF Braz. J. Chem, 2004 - SciELO Brasil	basic sites distribution. The amount of CO 2 chemisorbed and its desorption profile for each sample were measured on a <u>Micromeritics</u> <u>2900</u> TPR/TPD analyzer with a quadrupole <u>mass</u> <u>spectrometer</u> detector. Prior to analysis	2900
Infrared spectroscopy, thermoprogrammed desorption, and nuclear magnetic	A Corma, C Corell, V Fornes, W Kolodziejski, J Pérez Zeolites, 1995	cm-1) were scaledaccording to the sample weight. Temperature programmed desorption experiments were done in a <u>Micromeritics 2900</u> apparatus. The Unfortunately, our tpd, apparatusdoes not work on line with a <u>mass spectrometer</u> , so wehave not been able to clarify	2900
Initiation step and reactive intermediates in the transformation of methanol into	AT Aguayo, AG Gayubo, R Vivanco, A Alonso, J Ind. Eng. Chem, 2005	The runs of injection of methanol (or methanol and water) pulses have been carried out in an AutoChem II (Micromeritics) adsorption–desorption device connected on-line (by means of a thermostated line) to a <u>mass</u> <u>spectrometer</u> (Balzers Instruments)	2920
Inverse temperature dependence due to catalyst deactivation in liquid phase citral hydrogenation over Pt/AI2O3	Maki-Arvela, P., Kumar, N., Eranen, K., Salmi, T., Murzin, D.Yu., Chemical Engineering Journal, 122 (3), p.127-134, Sep 2006	min and temperature programme 10K/min up to 923K (30min) by using <u>Micromeritics</u> (<u>AutoChem 2910</u>) apparatus and analyzing the desorbing gases by a quadrupole <u>mass spectrometer</u> (Balzers Instrument, Omnistar). In the temperature programmed oxidation	2910
Investigation of the catalytic performances of supported noble metal based catalysts in the NO+H2 reaction under lean	Engelmann-Pirez, M., Granger, P., Leclercq, G., Catalysis Today, 107, p.315-322, Oct 2005	Temperature-programmed reduction (TPR) was carried out in a <u>Micromeritics AutoChem II 2920</u> . Surface compositions were obtained by4000h-1. The effluents were analysed by a Balzers <u>mass spectrometer</u> and a HP 5890 series II chromatograph fitted	2920
InVO4-sensitized TiO2 photocatalysts for efficient air purification with visible light	G Xiao, X Wang, D Li, X Fu - Journal of Photochemistry & Photobiology, A:, 2008	volume of the samples were collected at 77 K using <u>Micromeritics</u> ASAP 2010 of oxygen (O 2 TPD) using a volumetric flow apparatus (<u>AutoChem 2910</u> , Micrometrics of the desorbed gases were performed continuously with a quadrupole <u>mass spectrometer</u> (Balzers OminiStar	2910

IR investigation of the interaction of deuterium with Ce0.6Zr0.4O2 and Cl-doped Ce0.6Zr0.4O2	Gennari, F.C., Montini, T., Hickey, N., Fornasiero, P., Graziani, M., Applied Surface Science, 252 (24), p.8456-8465, Oct 2006	were obtained on a <u>Micromeritics</u> ASAP 2000 analyzerSensorlab quadropole <u>mass</u> <u>spectrometer</u> . Typically, 0.2gcapillary tube to the <u>mass spectrometer</u> . Analysis was performedexperiments with a <u>Micromeritics</u> TPD/TPR <u>2900</u> apparatus. After	2900
Iridium-supported catalyst for enantioselective hydrogenation of 1-phenyl-1,2-propanedione: The effects of the addition	Marzialetti, T., Fierro, J.L.G., Reyes, P., Catalysis Today, 107, p.235- 243, Oct 2005	using a gas chromatograph- <u>mass spectrometer</u> (GCMS-QP5050 Shimadzu) providedwas studied in a TPR/TPD <u>2900 Micromeritics</u> system equipped with a thermalof N2at 77K in an automatic <u>Micromeritics</u> system Model ASAP 2010. The	2900
Iron-catalyzed propylene epoxidation by nitrous oxide: Effect of boron on structure	S Yang, W Zhu, Q Zhang, Y Wang - Journal of Catalysis, 2008	H 2 temperature-programmed reduction (H 2 - TPR), NH 3 temperature-programmed desorption (NH 3 -TPD), and CO 2 -TPD were performed using a <u>Micromeritics</u> <u>AutoChem II 2920</u> instrument connected to a ThermoStar GSD 301 T2 <u>mass spectrometer</u>	2920
Iron-catalyzed propylene epoxidation by nitrous oxide: Studies on the effects of	X Wang, Q Zhang, S Yang, Y Wang - J. Phys. Chem. B, 2005	NH 3 -temperature-programmed desorption (NH 3 -TPD) was also carried out with the <u>Micromeritics</u> <u>AutoChem II 2920</u> equipment, which was connected with a ThermoStar GSD 301 T2 <u>mass</u> <u>spectrometer</u> (Pfeiffer Vacuum)	2920
Kinetically controlled synthesis of carbon nanofibers with different morphologies by catalytic CO disproportionation	Lu, WX., Sui, ZJ., Zhou, JH., Li, P., Chen, D., Zhou, X G., Chemical Engineering Science, 65 (1), p.193-200, Jan 2010	by an on-line <u>mass spectrometer</u> and the morphologies308 to 1223K on <u>Micromeritics</u> <u>AutoChem II 2920</u> . N2adsorptioncarried out with <u>Micromeritics</u> ASAP 2010. Scanline quadruple <u>mass spectrometer</u> (Questor, ABB	2920
Kinetics and stereoselectivity of o- xylene hydrogenation over Pd/Al2O3	AK Neyestanaki, P Mäki-Arvela, H Backman, H Journal of Molecular , 2003	Auto Chem 2910 , Micromeritics). Prior to the adsorption studies, the samples were reduced in situ at 673 K. For the H 2 -TPD studies, the hydrogen adsorption was carried out at 363 K. Desorbed gases were identified and analysed by a quadruple <u>mass spectrometer</u> (Omnistar	2910
Kinetics and thermodynamics of the Cr (III) adsorption on the activated carbon from	SI Lyubchik, AI Lyubchik, OL Galushko, LP Colloids and Surfaces A:, 2004	The carbon surfaces were characterized by Boehm titration methods and temperature-programmed desorption with <u>Micromeritics</u> TPD/TPR <u>2900</u> instrument using a quartz microreactor, which was connected to a <u>mass</u> <u>spectrometer</u> set-up (Fisons MD800) for continuous	2900
Kinetics and thermodynamics of the Cr(III) adsorption on the activated carbon from comingled wastes	Lyubchik, S.I., Lyubchik, A.I., Galushko, O.L., Tikhonova, L.P., Vital, J., Fonseca, I.M., Lyubchik, S.B., Colloids and Surfaces A: Physicochemical and Engineering Aspects, 242 (1), p.151-158, Aug 2004	Area & Porosimetry Analyzer, <u>Micromeritics</u> ASAP 2010 (Table 1 Tabletemperature- programmed desorption with <u>Micromeritics</u> TPD/TPR <u>2900</u> instrument using a quartzwhich was connected to a <u>mass spectrometer</u> set-up (Fisons MD800) for	2900

Kinetics of H2 recovery from dodecahydro-N-ethylcarbazole over a supported Pd	F Sotoodeh, L Zhao, KJ Smith - Applied Catalysis A, General, 2009	was determined by pulsed chemisorption using the same Micromeritics AutoChem II 2920 A Micromeritics ASAP 2020 Accelerated Surface Area and Porosimetry analyzer was gas composition was continuously monitored using a quadrupole mass spectrometer (SRC Residual	2920
Kinetics of the NO/H2 reaction on Pt/LaCoO3: A combined theoretical and 	F Dhainaut, S Pietrzyk, P Granger - Journal of Catalysis, 2008	H 2 temperature-programmed reduction (TPR) experiments were carried out on a <u>Micromeritics</u> <u>AutoChem II 2920</u> instrument under a flow of The gaseous mixture was analysed using a Balzers quadrupole <u>mass spectrometer</u> and a Hewlett Packard 5890 series II chromatograph	2920
Kinetics of the NO+ H2 reaction over supported noble metal based catalysts:	F Dhainaut, S Pietrzyk, P Granger - Applied Catalysis B, Environmental, 2007	programmed reduction and hydrogen titration measurements, performed at 110 °C [9], were carried out on a Micromeritics AutoChem II 2920 The gaseous mixture was analysed by a Balzer mass spectrometer and a Hewlett Packart 5890 series II chromatograph fitted with a	2920
Kinetics of the NO+H2 reaction over supported noble metal based catalysts: Support effect on their adsorption	Dhainaut, F., Pietrzyk, S., Granger, P., Applied Catalysis B, Environmental, 70 (1), p.100-110, Jan 2007	performed at 110^oC[9], were carried out on a Micromeritics AutoChem II 2920 using a pulse technique. Prior to chemisorptionThe gaseous mixture was analysed by a Balzer mass spectrometer and a Hewlett Packart 5890 series II chromatograph	2920
Kinetics of Water-Gas-Shift- Reaction using MoS 2 catalyst dotted with Cofzk.de	HJ Ederer, T Fritsch, E Henrich, CE Mas - 2002 - fzk.de	The activated catalyst CoMo-C49 was analysed with a surface investiga- tion equipment <u>AutoChem 2910</u> from <u>Micromeritics</u> company The second analytic device was a quadrupol <u>mass spectrometer</u> of Pfeiffer company (Thermo Star 200) with 2 detectors (Faraday, CH	2910
Kinetics, catalyst deactivation and modeling in the hydrogenation of 2- sitosterol to 2-sitostanol over microporous and	Maki-Arvela, P., Martin, G., Simakova, I., Tokarev, A., Warna, J., Hemming, J., Holmbom, B., (), Murzin, D.Yu., Chemical Engineering Journal, 154 (1), p.45-51, Nov 2009	dispersion was measured by pulse CO-chemisorption (<u>Micromeritics AutoChem</u> 2901). The catalyst was prereduced at 100to 4bars. The raw material according to the <u>mass spectrometer</u> results contained: 82wt.% beta-sitosterol	2910
L13psu.edu	M RAZISKOVALCI, Y RESEARCHERS - Poroilo o delu 2005 Annual report 2005 - Citeseer	FIGURE: Automated system for heterogeneous catalysts char- acterization (<u>Micromeritics</u> , model <u>AutoChem II 2920</u>), connected to a <u>mass</u> <u>spectrometer</u> (Pfeiffer Vacuum, model Thermostar) employed as a second- ary detector. Page 155	2920
La, Ca and Fe oxide perovskites: preparation, characterization and catalytic properties for methane combustion	Ciambelli, P., Cimino, S., Lisi, L., Faticanti, M., Minelli, G., Pettiti, I., Porta, P., Applied Catalysis B: Environmental, 33 (3), p.193-203, Oct 2001	reduction (TPR) experiments were performed as reported in [9] using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyzer equipped with a TC detector and coupled with a Hiden HPR 20 <u>mass spectrometer</u> . Samples (100 mg) were preheated in flowing air at 1073 K for 2	2900

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Liquid-Impregnated Clay Solid Sorbents for CO Removal from Postcombustion Gas	R Siriwardane, C Robinson - Journal of Environmental Engineering, 2009 - link.aip.org	reactor (<u>Micromeritics AutoChem 2910</u> , Norcross, Ga.) with the simulated flue–gas mix with moisture (by volume) at at a flow rate of; they were regenerated at with steam/nitrogen. The outlet gaseous mixture was analyzed by a Pfeiffer Vacuum Thermostar <u>mass spectrometer</u>	2910
Liquid-phase hydroisomerization of n- octane over platinum- containing zeolite	A de Lucas, P Sanchez, A Funez, MJ Ramos, JL Ind. Eng. Chem, 2006	of the acid sites was measured by temperature programmed desorption of ammonia (TPDA) using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyzer Liquid products were analyzed in a gas chromatograph (GC-17A SHIMADZU) coupled to a <u>mass</u> <u>spectrometer</u> (QP-5000 SHIMADZU)	2900
Low temperature carbon monoxide oxidation over gold nanoparticles supported on sodium titanate nanotubes	JY Tsai, JH Chao, CH Lin - Journal of Molecular Catalysis. A, Chemical, 2009	of the catalytic reactivity was performed in an AutoChem 2910 analyzer (Micromeritics) equipped with The AutoChem 2910 was operated in a pulse reactor mode, and its diagram The product mixture were analysed by an on-line quadrupole mass spectrometer (Prolab, Thermo	2910
Low temperature decomposition of nitrous oxide over Fe/ZSM-5: Modelling of theepfl.ch	L Kiwi-Minsker, DA Bulushev, A Renken - Catalysis Today, 2005	reactivity of deposited oxygen and temperature-programmed desorption/reaction (TPD/TPR) experiments were performed in a <u>Micromeritics</u> <u>AutoChem 2910</u> analyser provided with a quartz plug-flow reactor. A ThermoStar 200 (Pfeiffer Vacuum) mass-spectrometer was used to	2910
Low Temperature Water– Gas Shift/Methanol Steam Reforming: Alkali Doping to 	HN Evin, G Jacobs, J Ruiz-Martinez, UM Graham, A Catalysis Letters, 2008	treated in-situ in a Micromeritics AutoChem II 2920 chemisorption analyzer under the following conditions: the samples were first reduced at 300 °C for 8 h CO2 desorbed was measured by a Pfeiffer/Balzers Thermostar mass spectrometer coupled to the Micromeritics system	2920
Low Temperature Water– Gas Shift: Alkali Doping to Facilitate Formate C–H Bond	HN Evin, G Jacobs, J Ruiz-Martinez, GA Thomas, BH Catalysis Letters, 2008	The catalysts were treated in-situ in a Micromeritics AutoChem II 2920 chemisorption analyzer under the following conditions: the samples were first reduced at 300 °C for 8 h was measured by a Pfeiffer/Balzers Thermostar mass spectrometer coupled to the Micromeritics system	2920
Low-temperature catalytic decomposition of N 2 O on platinum and bismuthrsc.org	R Burch, GA Attard, ST Daniells, DJ Jenkins, JP Chemical, 2002	Reaction products were monitored using a computer interfaced Fisons Gaslab 300 <u>Mass spectrometer</u> , operated using the corresponding Thermosoft at 300 °C. Characterisation of the catalysts was achieved using H2 and CO chemisorption (<u>Micromeritics</u> <u>AutoChem</u> <u>2910</u>) and	2910
Low-Temperature Single- Wall Carbon Nanotubes Synthesis: Feedstock polyu.edu.hk	E Mora, JM Pigos, F Ding, BI Yakobson, AR Journal of the, 2008	by an impregnation method.(8) The catalyst reducibility was studied by temperature programmed reduction (TPR) in a <u>Micromeritics AutoChem</u> 2910 under 10% H A <u>mass spectrometer</u> (MS), attached at the gas outlet of the reactor, monitored the catalyst activity in situ during the	2910

Manganese oxide catalysts supported on TiO2, Al2O3, and SiO2: a comparison for	PG Smirniotis, PM Sreekanth, DA Pena, RG Ind. Eng. Chem, 2006	The temperature-programmed reduction experiments were carried out from 353 to 1223 K on a <u>Micromeritics AutoChem</u> 2910 instrument using 50 mg of oxygen for 2 h at 673 K. The reactants and products were analyzed online using a Quadrapole <u>mass spectrometer</u> (MKS PPT	2910
Manganese Oxide/Titania Materials for Removal of NO x and Elemental Mercury	L Ji, PM Sreekanth, PG Smirniotis, SW Thiel, NG Energy & Fuels, 2008	programmed reduction (H 2 -TPR) experiments were carried out from 80-950 °C using a <u>Micromeritics AutoChem 2910</u> instrument The reactants and products were analyzed online using a Quadrapole <u>mass spectrometer</u> (MKS PPT-RGA), chemiluminescence detector (Eco	2910
Manganese-promoted Rh/Al2O3 for C2- oxygenates synthesis from syngas - Effect of manganese loading	Ojeda, M., Granados, M.L., Rojas, S., Terreros, P., Garcia-Garcia, F.J., Fierro, J.L.G., Applied Catalysis A: General, 261 (1), p.47-55, Apr 2004	experiments were carried out on a <u>Micromeritics</u> TPD/TPR <u>2900</u> apparatus. The catalystPrisma QMS 200 TM quadropole <u>mass spectrometer</u> . The catalyst was reducedreactor connected to the <u>mass spectrometer</u> as described above. The	2900
MCM-41 supported Mo/Zr mixed oxides as catalysts in liquid phase condensation of 2- methylfuran with acetone	Li, T., Cheng, S., Lee, JF., Jang, L Y., Journal of Molecular Catalysis A: Chemical, 198 (1), p.139-149, May 2003	nitrogen temperature with a <u>Micromeritics</u> ASAP 2000 apparatus. TransmissionTPD) was carried out on a <u>Micromeritics</u> <u>AutoChem</u> <u>2910</u> instrument. 50 mg ofmonitored by a quadruple <u>mass spectrometer</u> (Thermo ONIX, ProLab) connected	2910
Mesoporous silica- aluminas derived from precipitation: a study of the acidity,	M Bartoszek, R Eckelt, C Jäger, H Kosslick, A Pawlik - Journal of Materials 	Measurements were carried on an AutoChem 2910 (Micromeritics) apparatus that was coupled with a mass spectrometer to differentiate between ammonia and water desorption. Approximately, 100 mg of the sample was activated in flowing helium at elevated temperature	2910
Methanation of carbon dioxide over the LaNiO3 perovskite catalysts activated	J GAO Journal of Fuel Chemistry and Technology, 2009	The sample was deposited on a Cu grid for TEM observation. H 2 -temperature- programmed desorption (H 2 -TPD) was performed on <u>Micromeritics AutoChem 2920</u> II instrument connected to a ThermoStar GSD 301 T2 <u>mass spectrometer</u>	2920
Methane combustion and CO oxidation on LaAl1-xMnxO3 perovskite-type oxide	S Cimino, L Lisi, S De Rossi, M Faticanti, P Applied Catalysis B, , 2003	Temperature programmed reduction (TPR) experiments were performed using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyser equipped with a TC detector and coupled with a Hiden HPR 200 <u>mass</u> <u>spectrometer</u> . Samples (0.1 g) were	2900
Microwave plasma assisted preparation of Pd-nanoparticles with controlled dispersion on woven activated carbon fibres	Korovchenko, P., Renken, A., Kiwi- Minsker, L., Catalysis Today, 102, p.133-141, May 2005	for characterization of surface acidity using a Micromeritics AutoChem 2910 analyzer. For the TPD measurements 0.05ganalyzed by ThermoStar-200 quadrupole "on-line" mass- spectrometer (Pfeiffer Vacuum) calibrated with gas mixtures	2910
Mild hydrogenation of quinoline - 2. A novel Rh- containing pillared layered clay catalyst	Campanati, M., Casagrande, M., Fagiolino, I., Lenarda, M., Storaro, L., Battagliarin, M., Vaccari, A., Journal of Molecular Catalysis A: Chemical, 184 (1), p.267-272, Jun 2002	carried out at 77 K using a Micromeritics ASAP 2010. Before each measurementmeasurements were performed using a Micromeritics AutoChem 2910. High resolution TEMthickness 0.25 mu m) and mass spectrometer detector. 3 Results and discussion	2910

Mild hydrogenation of quinoline 2. A novel Rh-containing pillared layered clay	M Campanati, M Casagrande, I Fagiolino, M Journal of Molecular , 2002	Hydrogen chemisorption measurements were performed using a <u>Micromeritics AutoChem 2910</u> were carried out using a GC–MS Hewlett-Packard GCD 1800A system equipped with an HP-5 column (30 m×0.25 mm, film thickness 0.25 µm) and <u>mass spectrometer</u> detector	2910
Modification of polystyrene- based activated carbon spheres to improve adsorption	Q Wang, X Liang, W Qiao, C Liu, X Liu, R Zhang, Applied Surface, 2009	acidity measurement [35]. TPD experiment was carried out on an AutoChem II 2920 (Micromeritics , USA). A typical procedure 10 °C/min. The outlet gas was diverted to a quadrupole mass spectrometer (Questor, ABB Extrel, USA) for analysis	2920
Modification of the adsorption properties of high surface area graphites by oxygen functional groups	Cuervo, M.R., Asedegbega-Nieto, E., Diaz, E., Ordonez, S., Vega, A., Dongil, A.B., Rodriguez-Ramos, I., Carbon, 46 (15), p.2096-2106, Dec 2008	at -196^oC with a <u>Micromeritics</u> ASAP 2000 surfaceTPO), employing a <u>Micromeritics</u> TPD- <u>2900</u> apparatus connectedPfeiffer Vacuum-300 <u>mass spectrometer</u> . For this purposewith a quadrupole <u>mass spectrometer</u> (Balzers QMG 421-C	2900
Monodispersed Pd Nanoparticles for Acetylene Selective Hydrogenation: epfl.ch	M Ruta, N Semagina, L Kiwi - The Journal of Physical, 2008 - infoscience.epfl.ch	decom- position (TPD) in He (20 mL/min, ramp rate 20 K/min from room temperature up to 1273 K) using a <u>Micromeritics AutoChem 2910</u> analyzer. The amounts of CO and CO 2 desorbed were monitored with a ThermoStar-200 quadropole <u>mass spectrometer</u> (Pfeiffer Vacuum	2910
Monolithic Pt/Ce0.8Zr0.2O2/cordierite catalysts for low temperature water gas shift reaction in the real	Du, X., Gao, D., Yuan, Z., Liu, N., Zhang, C., Wang, S., International Journal of Hydrogen Energy, 33 (14), p.3710-3718, Jul 2008	chemisorption at 40^oC[21]was performed on Micromeritics AutoChem 2920 equipment, assuming that eachPt dispersion measurements, using Micromeritics AutoChem 2920 equipped with a quantum mass spectrometer (OmniStar). The sample was heated	2920
Morphology observation of carbon deposition by CH4 decomposition over Ni- based catalysts	Yonglai Yang, Hengyong Xu, Wenzhao Li , Nanotechnology, 16 (1), p.129-132, Jan 2005	oxidation behaviour of the carbon de- posits on a Micromeritics AutoChem 2910 system. The catalysts after carbon depositiondeposits, were detected by an on-line Omnistar mass spectrometer. 3. Results and discussion The TEM images in	2910
N2O decomposition over K- promoted Co-Al catalysts prepared from hydrotalcite- like precursors	Cheng, H., Huang, Y., Wang, A., Li, L., Wang, X., Zhang, T., Applied Catalysis B, Environmental, 89 (3), p.391-397, Jul 2009	N2adsorption at -196^oC using a Micromeritics ASAP 2010 apparatus. The X-rexperiments were carried out on a Micromeritics AutoChem II 2920 automated catalystcharacterizationwas monitored online by a mass spectrometer (Omini-star, GSD-300), withm	2920
Nanostructured CuxCe1- xO2- y mixed oxide catalysts: Characterization and WGS	A Pintar, J Batista, S Hočevar - Journal of colloid and interface science, 2007	TPO, TPD-H 2 , and selective N 2 O pulse reaction measurements were performed using an automated <u>Micromeritics AutoChem II 2920</u> samples CuCe-1 and CuCe-4) was done by TCD and MS detectors (computer-interfaced Pfeiffer Vacuum ThermoStar <u>mass spectrometer</u>)	2920
NEW Ni/MIEC CERMET ANODE FOR SOFC APPLICATIONS BASED ON inpl-nancy.fr	F MOSER, MT CALDES, O JOUBERT, V GARCIA perso.ensem.inpl- nancy.fr	The temperature-programmed reduction (TPR) studies were performed in a chemisorption unit Micromeritics AutoChem 2910 with powder samples of 50 mg The composition of the reactants / products mixture was analysed with an on-line mass spectrometer	2910

J Requies, MA Cabrero, VL Barrio, JF Cambra, MB Catalysis Today, 2006	A <u>Micromeritics</u> TPD/TPR <u>2900</u> apparatus equipped with a TCD was used for temperature-programmed reduction (TPR) analyses on catalyst samples (50 mg) loaded in a U-shaped quartz reactor connected to a Baltzer Prisma QMS 200 TM quadrupole <u>mass spectrometer</u>	2900
PH Mutin, AF Popa, A Vioux, G Delahay, B Applied Catalysis B,, 2006	of ammonia (NH 3 -TPD) and temperature-programmed reduction by H 2 (H 2 -TPR) were performed with a Micromeritics AutoChem 2910 apparatus O 2 , N 2 , H 2 O, and N 2 O were continuously monitored by on-line sampling to a quadrupole <u>mass spectrometer</u> (Balzers QMS	2910
AF Popa, PH Mutin, A Vioux, G Delahay, B Chemical Communications, 2004	was analysed by nitrogen physisorption at 77 K on a <u>Micromeritics</u> ASAP 2000 Temperature programmed desorption (TPD) of NH3 (Micromeritics <u>AutoChem</u> 2910) showed that the tinuously monitored by on-line sampling to a quadrupole <u>mass spectrometer</u> (Balzers QMS 421	2910
RV Siriwardane, C Robinson, M Shen, T Simonyi - Energy Fuels, 2007	Competitive gas adsorption and desorption studies were conducted in a lab-scale fixed-bed reactor (Micromeritics AutoChem 2910 atmospheric flow reactor) at 14.7 psi (1.01 × 10 5 Pa The outlet gas stream was analyzed using a Pfeiffer Vacuum Thermostar mass spectrometer	2910
KM NIKOLAJSEN - biblion.epfl.ch	Micromeritics Reactor	
Datta, A., Sakthivel, S., Kaur, M., Venezia, A.M., Pantaleo, G., Longo, A., Microporous and Mesoporous Materials, 128 (1), p.213-222, Mar 2010	reaction (TPR) measurements were conducted with a <u>Micromeritics AutoChem</u> <u>2910</u> Automated Catalyst Characterization Systemthe evolved gases were analysed by a quadrupole <u>mass</u> <u>spectrometer</u> . The average oxidation state of vanadium was	2910
Li, T., Wong, ST., Chao, MC., Lin, H P., Mou, CY., Cheng, S., Applied Catalysis A: General, 261 (2), p.211-219, Apr 2004	nitrogen temperature with a <u>Micromeritics</u> ASAP 2000 apparatus. TransmissionTPD) was carried out on a <u>Micromeritics AutoChem 2910</u> instrument. TPD profilesmonitored by a quadruple <u>mass spectrometer</u> (Thermo ONIX, ProLab) connected	2910
W Wang, JH Wang, CL Chen, NP Xu, CY Mou - Catalysis Today, 2004	The TPR and NH 3 -TPD were carried out on a Micromeritics AutoChem 2910 instrument The desorption process was monitored by a Quadruple Mass spectrometer (Thermo ONIX, ProLab) connected on-line through a heated capillary interface	2910
JN Kuhn, Z Zhao, LG Felix, RB Slimane, CW Applied Catalysis B, , 2008	The effluent was monitored with a Cirrus RGA mass spectrometer using the Faraday detector Temperature-programmed reduction (TPR) and oxidation (TPO) were performed with a Micromeritics AutoChem II 2920 equipped with a TCD	2920
	Cabrero, VL Barrio, JF Cambra, MB Catalysis Today, 2006 PH Mutin, AF Popa, A Vioux, G Delahay, B Applied Catalysis B,, 2006 AF Popa, PH Mutin, A Vioux, G Delahay, B Chemical Communications, 2004 RV Siriwardane, C Robinson, M Shen, T Simonyi - Energy Fuels, 2007 KM NIKOLAJSEN - biblion.epfl.ch Datta, A., Sakthivel, S., Kaur, M., Venezia, A.M., Pantaleo, G., Longo, A. , Microporous and Mesoporous Materials, 128 (1), p.213-222, Mar 2010 Li, T., Wong, ST., Chao, MC., Lin, HP., Mou, CY., Cheng, S. , Applied Catalysis A: General, 261 (2), p.211-219, Apr 2004 W Wang, JH Wang, CL Chen, NP Xu, CY Mou - Catalysis Today, 2004 JN Kuhn, Z Zhao, LG Felix, RB Slimane, CW Applied Catalysis B,	Cabrero, VL Barrio, JF Cambra, MB Catalysis Today, 2006 PH Mutin, AF Popa, A Vioux, G Delahay, B Applied Catalysis B,, 2006 PH Mutin, AF Popa, A Vioux, G Delahay, B Applied Catalysis B,, 2006 A Vioux, G Delahay, B Applied Catalysis B,, 2006 A F Popa, PH Mutin, A Vioux, G Delahay, B Chemical Communications, 2004 RV Siriwardane, C Robinson, M Shen, T Simonyi - Energy Fuels, 2007 KM NIKOLAJSEN-biblion-epfl.ch WM NIKOLAJSEN-biblion-epfl.ch WM NIKOLAJSEN-biblion-epfl.ch WM NIKOLAJSEN-biblion-epfl.ch WM NIKOLAJSEN-biblion-epfl.ch Datta, A., Sakthivel, S., Kaur, M., Pantaleo, G., Longo, A., Microporous and Mesoporous Materials, 128 (1), p.213-222, Mar 2010 Li, T., Wong, ST., Chao, MC., Lin, HP., Mou, C.Y., Cheng, S., Applied Catalysis Robert Catalysis Today, 2004 WW ang, JH Wang, CL Chen, NP Xu, Cry Mou - Catalysis Today, 2004 WW ang, JH Wang, CL Chen, NP Xu, Cry Mou - Catalysis Today, 2004 JN Kuhn, Z Zhao, LG Felix, RB Slimane, CW Applied Catalysis B, 2008 JN Kuhn, Z Zhao, LG G Felix, RB Slimane, CW Applied Catalysis B, 2008 Micromeritics Autochem 2910 instrument. The desorption process was monitored with a Micromeritics and connected on line through a heated capiliary interface Line programmed freduction by H 2 (H 2 - TPR) were carried out on a Micromeritic source with a Micromeritic source with a Catalysis And Chem 2910 instrument. The desorption process was monitored by a quadruple Mass spectrometer (Thermo ONIX, ProLab) connected on line through a heated capillary interface JN Kuhn, Z Zhao, LG Felix, RB Slimane, CW Applied Catalysis B applied Catalysis

On the catalytic nature of Mn/sulfated zirconia for selective reduction of NO withdicp.ac.cn	N Li, A Wang, Z Liu, X Wang, M Zheng, Y Applied Catalysis B,, 2006	Temperature programmed reduction (TPR) experiments were carried out with <u>Micromeritics AutoChem II 2920</u> Automated Catalyst Characterization System using an H 2 NO-TPD was conducted on a flowing reaction system using a	2920
One-pot synthesis and characterization of metal phosphide-doped carbon xerogels	H Wang, Y Shu, A Wang, J Wang, M Zheng, X Wang, T Carbon, 2008	 mass spectrometer (omini-star, GSD-300) as The carbothermal reduction process of the organic xerogels was monitored by running the temperature-programmed reaction on a Micromeritics AutoChem 2910 apparatus combining with an Ominatar mass spectrometer. Before 	2910
Oxidation of pinane using transition metal acetylacetonate complexes immobilised on modified activated carbon	Valente, A., Botelho do Rego, A.M., Reis, M.J., Silva, I.F., Ramos, A.M., Vital, J., Applied Catalysis A: General, 207 (1), p.221-228, Feb 2001	nitrogen at 77 K on a Micromeritics ASAP 2010 V1.01 BMICROMERITCS TPD/TPR 2900 instrument by heatingstream was fed to a mass spectrometer through a heated linefragmentation in the mass spectrometer, of imine 3 which	2900
Oxidation of propane to acrylic acid over vanadyl pyrophosphate: modifications of the structural and acid properties	Landi, G., Lisi, L., Volta, JC., Journal of Molecular Catalysis A: Chemical, 222 (1), p.175-181, Nov 2004	TPD) was performed using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyser equipped with acoupled with a Hiden HPR 20 <u>mass spectrometer</u> . After a pre-treatment atoxides were followed with the <u>mass spectrometer</u> . In our experiments only	2900
Oxidative dehydrogenation of ethane on γ-Al2O3 supported vanadyl and iron	MP Casaletto, L Lisi, G Mattogno, P Patrono, G Applied Catalysis A, , 2002	TPR with hydrogen and TPD of NH 3 were carried out using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyser equipped with a TC detector and coupled with a Hiden HPR 20 <u>mass spectrometer</u> . In the TPR experiments, the sample was	2900
Oxidative dehydrogenation of propane over catalysts based on carbon nanofibers	Z Sui, J Zhou, Y Dai, W Yuan - Catalysis Today, 2005	creating products. TPD and TPSR runs were carried out on AutoChem II 2920 (Micromeritics , USA). The outgoing gas was diverted to a quadrupole <u>mass spectrometer</u> (Questor, ABB Extrel, USA) to be analyzed. Concentrations	2920
Oxygen Exchange Kinetics over Sr-and Co-Doped LaFeO3	JN Kuhn, PH Matter, JMM Millet, RB Watson, The Journal of, 2008	Temperature-Programmed Reoxidation Experiments were performed with an AutoChem II 2920 were performed by monitoring the effluent with a gas chromatograph/mass spectrometer (GC/MS A Micromeritics ASAP 2010 instrument was used to treat samples under various	2920
Palladium based catalysts for exhaust aftertreatment of natural gas powered vehicles and biofuel combustion	Klingstedt, F., Neyestanaki, A.K., Byggningsbacka, R., Lindfors, LE., Lunden, M., Petersson, M., Tengstrom, P., (), Vayrynen, J., Applied Catalysis A: General, 209 (1), p.301-316, Feb 2001	detected using a quadrupole <u>mass spectrometer</u> (Carlo Erba Instruments600 C (10 C/min) using a <u>Micromeritics AutoChem 2910</u> equipped with a thermaldetected using a quadruple <u>mass spectrometer</u> . The surface composition	2910
Palladium Catalysts Supported on Fishbone Carbon Nanofibers from Different	J ZHOU, Z SUI, X ZHOU, W YUAN - Chinese Journal of Catalysis, 2008	The temperature-programmed desorption (TPD) experiments were carried out on an Auto- chem II 2920 (Micromeritics, USA) combined with a quadrupole mass spectrometer (Questor, ABB Extrel, USA). De- tailed procedures refer to the literature [10]	2920

Palladium on carbon nanofibers grown on metallic filters as novel structuredepfl.ch	P Tribolet, L Kiwi- Minsker - Catalysis today, 2005	min, ramp rate 10 K/min) using a Micromeritics AutoChem 2910 analyzer. In these experiments about 80 mg of 6% CNF/SMF Inconel were placed in a quartz tubular reactor. The TPD products were analyzed by a ThermoStar-200 quadrupole mass spectrometer (Pfeiffer Vacuum	2910
Partial oxidation of toluene to benzaldehyde and benzoic acid over model vanadia/epfl.ch	DA Bulushev, F Rainone, L Kiwi- Minsker - Catalysis Today, 2004	Temperature programmed reduction (TPR) experiments were performed in a <u>Micromeritics</u> <u>AutoChem 2910</u> analyser with a quartz plug-flow reactor. Hydrogen concentration was determined by a Thermostar 200 quadrupole mass-spectrometer (Pfeiffer Vacuum)	2910
Photocatalytic Generation of H 2 Gas from Neat Ethanol over Pt/TiO 2 Nanotube	CH Lin, CH Lee, JH Chao, CY Kuo, YC Cheng, WN Catalysis Letters, 2004	and 900 nm. TPD/NH3 experiments were performed with on a AutoChem 2910 automated catalyst characterization system (Micromeritics) interfaced with qudrapole <u>mass spectrometer</u> (Proleb, Thermo Onix). XRD spetra were	2910
Platinum catalysts on alumina and silica prepared by gas- and liquid- phase deposition in cinnamaldehyde hydrogenation	Lashdaf, M., Lahtinen, J., Lindblad, M., Venalainen, T., Krause, A.O.I., Applied Catalysis A: General, 276 (1), p.129-137, Nov 2004	determined with a VG 7070E high-resolution <u>mass</u> <u>spectrometer</u> (MS). For MS analysis the samples were heatedmethod. The experiments were carried out with a <u>Micromeritics</u> TPD/TPR <u>2910</u> <u>AutoChem</u> instrument. The sample (100-200 mg) was set	2910
Precursor Effect on the Molecular Structure, Reactivity, and Stability of Alumina	AE Lewandowska, MA Bañares, DF Khabibulin, OB 2009	and temperature-programmed oxidation (TPR/TPO) experiments were performed in a fixed-bed quartz reactor fitted to a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyzer 823 K; then it was cooled to 323 K. The TPR/TPO treatments were recorded with a Hiden HPR20 <u>mass spectrometer</u>	2900
Preparation and characterization of LaCrO3 and Cr2O3 methane combustion	MFM Zwinkels, O Haussner, P Govind Menon, SG Catalysis Today, 1999	Temperature-programmed reduction (TPR) of the powder samples was performed using a <u>Micromeritics</u> TPD/TPR <u>2900</u> , equipped with a thermal conductivity detector The combustion products were analyzed on-line using a Balzers QMG 421C quadrupole <u>mass spectrometer</u>	2900
Preparation of Cobalt Nitride from Co–Al Hydrotalcite and its Application indicp.ac.cn	H Cheng, Y Huang, A Wang, X Wang, T Zhang - Topics in Catalysis, 2009	Temperature programmed reduction (TPR) experiments were carried out on a <u>Micromeritics</u> <u>AutoChem II 2920</u> automated catalyst characterization system The outlet gas was monitored online by a <u>mass spectrometer</u> (Omini-star, GSD- 300), with an m/z of 16 representing NH3	2920
Preparation of Fischer- Tropsch cobalt catalysts supported on carbon nanofibers and silica using homogeneous	Bezemer, G.L., Radstake, P.B., Koot, V., van Dillen, A.J., Geus, J.W., de Jong, K.P., Journal of Catalysis, 237 (2), p.291-302, Jan 2006	Fisons Thermolab quadropole <u>mass spectrometer</u> through a capillary situatedwas executed with an AutoChem 2920 instrument from Micromeritics using a heating rate ofmeasurements were done with a Micromeritics ASAP 2010C. Before each	2920
Preparation of Fischer– Tropsch cobalt catalysts supported on carbon nanofibersuu.nl	GL Bezemer, PB Radstake, V Koot, AJ Van Dillen, Journal of Catalysis, 2006	The gases evolved were monitored by a Fisons Thermolab quadropole <u>mass spectrometer</u> through a capillary situated directly above the sample cup. Temperature-programmed reduction (TPR) was executed with an <u>AutoChem</u> <u>2920</u> instrument from <u>Micromeritics</u> using a	2920

Preparation, characterization and catalytic properties of carbon nanofiber- supported Pt, Pd, Ru monometallic particles	Taboada, C.D., Batista, J., Pintar, A., Levec, J., Applied Catalysis B, Environmental, 89 (3), p.375-382, Jul 2009	196^oC using a Micromeritics ASAP 2020 MPMicromeritics AutoChem II 2920 catalystPfeiffer Vacuum mass spectrometer (model ThermoStarout by TCD and mass spectrometer (MS) detectorsdetector and mass spectrometer (MS). We observed	2920
Pretreatments of Co3O4 at moderate temperature for CO oxidation at- 80° C	Y Yu, T Takei, H Ohashi, H He, X Zhang, M Haruta - Journal of Catalysis, 2009	O 2 -TPD and CO-TPR experiments were preformed at Automated Catalyst Characterization System (AutoChem 2920 , Micromeritics , USA) equipped with a <u>mass spectrometer</u> (QIC 20, Hiden, UK) by using 0.20 g catalyst powder	2920
Probing into the catalytic nature of Co/sulfated zirconia for selective reduction ofdicp.ac.cn	N Li, A Wang, M Zheng, X Wang, R Cheng, T Zhang - Journal of Catalysis, 2004	Temperature-programmed reduction (TPR) experiments were carried out with a <u>Micromeritics</u> <u>AutoChem II 2920</u> automated catalyst characterization system using an H 2 NO-TPD was conducted on a flowing reaction system using a <u>mass spectrometer</u> (Omini-star, GSD-300) as	2920
Promotional effects of noble metal addition to cobalt Fischer-Tropsch synthesis	D Xu, P Dai, Q Guo, W Li - Reaction Kinetics and Catalysis Letters, 2008	experiments of adsorbed CO were performed using a <u>Micromeritics AutoChem 2910</u> instrument at atmospheric pressure. The Page 3. DONGYAN XU et al.: NOBLE METAL 369 fragments mfe 2, 16, 18, 28 and 44 were recorded by a quadrupole <u>mass spectrometer</u> (Ominister 300	2910
Properties of alkali- promoted Cu-MgO catalysts and their activity for methanol	S Goodarznia, KJ Smith - Journal of Molecular Catalysis A: Chemical, 2010	A TCD attached to a Micromeritics AutoChem II chemisorption analyzer was used to detect the consumption of N 2 O and Cu dispersion was The gas flow lines between the pre-heater and the reactor as well between the reactor and the <u>mass spectrometer</u> were held at the same	2920
Pt/H-ZSM-12 as a catalyst for the hydroisomerization of C5-C7 n-alkanes and simultaneous saturation of benzene	Gopal, S., Smirniotis, P.G., Applied Catalysis A: General, 247 (1), p.113-123, Jul 2003	determined by pulse hydrogen chemisorption using a <u>Micromeritics AutoChem 2910</u> automated catalyst characterization systemHewlett-Packard, 5890 Series II) equipped with a <u>mass spectrometer</u> (Hewlett-Packard, 5972 Series II). Separation	2910
Pulse-response TAP studies of the reverse water-gas shift reaction over a Pt/CeO2 catalyst	Goguet, A., Shekhtman, S.O., Burch, R., Hardacre, C., Meunier, F.C., Yablonsky, G.S., Journal of Catalysis, 237 (1), p.102-110, Jan 2006	measured by BET (<u>Micromeritics</u> ASAP 2010) andH2chemisorption (<u>Micromeritics</u> <u>AutoChem</u> <u>2910</u>) and was found100C quadrupole <u>mass</u> <u>spectrometer</u> . The temperatureextracted from the <u>mass spectrometer</u> data using standard	2910
Quantification of Bronsted Acid Sites in Microporous Catalysts by a Combined FTIR	GVA Martins, G Berlier, C Bisio, S Coluccia, HO 2008	tubular furnace. The measurement was carried out on the activated catalysts using a Micromeritics TPD/TPR 2900 analyzer equipped with a TC detector and coupled with a Hiden HPR 20 <u>mass</u> <u>spectrometer</u> . The activated samples	2900
Quantification of metallic area of high dispersed copper on ZSM-5 catalyst by TPD	MCN Amorim de Carvalho, FB Passos, M Catalysis, 2002	H 2 TPD experiment was performed using a <u>Micromeritics</u> TPD/TPR <u>2900</u> apparatus, equipped with a Baltzer Mass Quadrupole, where the sample The degree of reduction of copper was also determined from the TPR data, coupled to <u>mass spectrometer</u> , previous to the TPD	2900
Reaction performance and characterization of Co/Al 2 O 3 Fischer–Tropsch	D Xu, W Li, H Duan, Q Ge, H Xu - Catalysis Letters, 2005	Temperature programmed surface reaction (TPSR) of adsorbed CO TPSR experiments of adsorbed CO were performed using <u>Micromeritics AutoChem</u> <u>2910</u> instrument The fragments m/e=2, 16, 18, 28 and 44 were recorded by a quadru- pole <u>mass spectrometer</u> (Ominister 300	2910

Reactivity of LSCF perovskites	Scott, S.P., Mantzavinos, D., Hartley, A., Sahibzada, M., Metcalfe, I.S., Solid State Ionics, 152, p.777-781, Dec 2002	Experiments were performed using a <u>Micromeritics</u> TPR/TPD <u>2900</u> machine fitted with a thermoconductivityunits. A Spectra Microvision <u>mass spectrometer</u> was attached to analyseusing a Spectra Microvision <u>mass spectrometer</u> attached to the outlet gas	2900
Reducibility of catalyzed cerium–praseodymium mixed oxides	W Chun, GW Graham, JA Lupescu, RW McCabe, MM Catalysis Letters, 2006	Temperature-programmed reduction (TPR) measurements were performed with a <u>Micromeritics AutoChem II 2920</u> system C/s) while concentrations of the three hydrocarbon species exiting the sample were monitored with a chemical ionization <u>mass spectrometer</u> [8]. Although	2920
Reduction of sulfur dioxide by carbon monoxide over doped nanophase cerium kth.se	AEC Palmqvist, MFM Zwinkels, Y Zhang, SG Nanostructured, 1997	with a Micromeritics TPD TPR 2900 Chemisorption Analyzer. Aseries of activity tests was carried out with a nearly stoichiometric reactant gas mixture containing 3.20 CO and 1.65S02inHe. The composition of outlet gases were analyzed with a Balzers mass spectrometer model	2900
Regeneration of CuO-ZnO-Al2O3/[gamma]-Al2O3 catalyst in the direct synthesis of	I Sierra, J Ereña, AT Aguayo, JM Arandes, J Applied Catalysis B: , 2009	The metallic surface area, determined by N 2 O chemisorption (in a <u>Micromeritics AutoChem 2920</u> coupled to a Pfeiffer Omnistar <u>mass spectrometer</u>), is 11.7 m 2 (g of catalyst) –1 . The physical properties, measured by N 2 adsorption–desorption (<u>Micromeritics</u> ASAP 2000) are	2920
Regeneration of CuO-ZnO-Al2O3/±c-Al2O3 catalyst in the direct synthesis of dimethyl ether	Sierra, I., Erena, J., Aguayo, A.T., Arandes, J.M., Bilbao, J., Applied Catalysis B, Environmental, 94 (1), p.108-116, Feb 2010	chemisorption (in a <u>Micromeritics AutoChem</u> 2920 coupled to a Pfeiffer Omnistar <u>mass</u> <u>spectrometer</u>), is 11.7m2Pfeiffer Omnistar <u>mass</u> <u>spectrometer</u>)[40], a totalcoupled to a <u>mass</u> <u>spectrometer</u> (Thermostar from	2920
Role of adsorbed NO in N2O decomposition over iron-containing ZSM-5 catalysts	DA Bulushev, A Renken, L Kiwi- Minsker - J. Phys. Chem. B, 2006	Catalytic Activity Measurements. Catalytic activity measurements and temperature- programmed studies were performed with a <u>Micromeritics AutoChem 2910</u> analyzer. A ThermoStar 200 (Pfeiffer Vacuum) quadrupole <u>mass spectrometer</u> was used for gas analysis	2910
Role of potassium on the structure and activity of alumina-supported vanadium oxide catalysts for propane oxidative	Garcia Cortez, G., Fierro, J.L.G., Banares, M.A., Catalysis Today, 78 (1), p.219-228, Feb 2003	was measured with a <u>Micromeritics</u> ASAP-2000 apparatusapparatus model TPR/TPD- <u>2900</u> fitted with a TCDcoupled to a quadrupole <u>mass</u> <u>spectrometer</u> equipment, model Balzersmonitored with an on-line <u>mass spectrometer</u> . The reaction products	2900
Roles of chlorine in the CO hydrogenation to C2- oxygenates over Rh-Mn- Li/SiO2 catalysts	Jiang, D., Ding, Y., Pan, Z., Li, X., Jiao, G., Li, J., Chen, W., Luo, H., Applied Catalysis A, General, 331, p.70- 77, Jan 2007	performed on a Micromeritics AutoChem 2910 apparatuswith a quadrupole mass spectrometer (Balzers OmniStarz=28) with the mass spectrometer. 2.2.6 Temperature-programmedrecorded with the mass spectrometer. 2.3 Catalytic	2910
Selective catalytic oxidation of ammonia to nitrogen at low temperature on Pt/CuO/Al2O3	Olofsson, G., Reine Wallenberg, L., Andersson, A., Journal of Catalysis, 230 (1), p.1-13, Feb 2005	adsorption isotherm, recorded on a <u>Micromeritics</u> ASAP 2400 instrument at liquidmicroreactor connected to a UTI 100C <u>mass spectrometer</u> [20]. Fifty milligrams ofTPR) was performed on a <u>Micromeritics</u> TPD/TPR <u>2900</u> instrument. The temperature	2900

Selective catalytic reduction of nitric oxide with ammonia on copper (II) ion	W Arous, H Tounsi, S Djemel, A Ghorbel, G Catalysis, 2005	Temperature programmed reduction by hydrogen (H 2 -TPR) was carried out with a <u>Micromeritics</u> <u>AutoChem 2910</u> apparatus using H 2 (3 The gas composition was monitored by sampling on line with a quadruple <u>mass spectrometer</u> (Pfeiffer Omnistar), calibrated with standard	2910
Selective catalytic reduction of nitric oxide with ammonia on Fe-ZSM-5 catalysts prepared by different methods	G Delahay, D Valade, A Guzmán- Vargas, B Applied Catalysis B, , 2005	TPR by H 2 /Ar (3/97, vol.%/vol.%) was carried out with a Micromeritics AutoChem 2910 apparatus using TCD detection The effluent composition was monitored continuously and by sampling on line to a quadruple <u>mass spectrometer</u> (Pfeiffer Omnistar) equipped with	2910
Selective CO removal in a H2-rich stream over supported Ru catalysts for the polymer electrolyte membrane fuel cell	Kim, Y.H., Park, E.D., Lee, H.C., Lee, D., Applied Catalysis A, General, 366 (2), p.363-369, Sep 2009	conducted in an <u>AutoChem 2910</u> unit (<u>Micromeritics</u>) equipped withconducted in an <u>AutoChem 2910</u> unit (<u>Micromeritics</u> conducted in an <u>AutoChem 2910</u> unit (<u>Micromeritics</u> and an online <u>mass spectrometer</u> (QMS 200, Pfeiffer	2910
Selective oxidation of CO in hydrogen-rich stream over Cu–Ce catalyst promoted gwangju.ac.kr	J Won Park, J Hyeok Jeong, WL Yoon, CS Kim, International Journal of, 2005	CO-TPR) and temperature-programmed oxidation (TPO) techniques, using a Balzers GSD 300T <u>mass spectrometer</u> (CO, CO 2 H 2 -TPR investigation was performed on a conventional temperature programming system (<u>AutoChem</u> 2910, <u>Micromeritics</u>) equipped with a	2910
Selective reduction of NO with CO over titania supported transition metal oxide	PM Sreekanth, PG Smirniotis - Catalysis Letters, 2008	The ammonia TPD experiments were performed on a Micromeritics AutoChem 2910 instrument using 50 mg of catalyst The reactants and products were analyzed on- line using a Quadrapole <u>mass spectrometer</u> (MKS PPT- RGA), chemiluminescence detector (Eco Physics CLD	2910
Significant effect of acidity on catalytic behaviors of Cs-substituted polyoxometalates for oxidative dehydrogenation of	Sun, M., Zhang, J., Cao, C., Zhang, Q., Wang, Y., Wan, H., Applied Catalysis A, General, 349 (1), p.212-221, Oct 2008	and were carried out with a Micromeritics TriStar 3000 surface areaNH3-TPD) were performed on Micromeritics AutoChem 2920 II instrument. Typicallyby ThermoStar GSD 301 T2 <u>mass spectrometer</u> . Because the parent peak	2920
Simultaneous Removal of NOx and Mercury in Low Temperature Selective	NG Pinto, PG Smirniotis - 2006 - osti.gov	were carried out from 353–1223 K on a Micromeritics AutoChem 2910 instrument were performed on a Micromeritics Tristar 3000 Porosimeter (Micromeritics Instrument Corporation at 673 K. The products were analyzed on- line using a Quadrapole mass spectrometer (MKS PPT	2910
Single and combined deactivating effect of alkali metals and HCl on commercial	L Lisi, G Lasorella, S Malloggi, G Russo - Applied Catalysis B, , 2004	Temperature programmed desorption (TPD) of NH 3 was carried out using a Micromeritics TPD/TPR 2900 analyser equipped with a TC detector and coupled with a Hiden HPR 20 <u>mass</u> <u>spectrometer</u> . After a pre-treatment in	2900
Skeletal isomerization of unsaturated fatty acids: the role of mesopores in HBeta	S Zhang, ZC Zhang - Catalysis Letters, 2007	The mea- surement was conducted on a Micromeritics AutoChem 2910 system from room temperature with a temperature ramp of 10 °C/min to 700 °C. Desorbed species were monitored with a calibrated thermoconductivity detector (TCD) as well as a mass spectrometer	2910
Sol-gel derived mesoporous Cr/Al 2 O 3 catalysts for SCR of NO by ammonia	F Ayari, M Mhamdi, G Delahay, A Ghorbel - Journal of Porous Materials	temperature ramped from ambient to 1,000 °C at 15 °C/min) was performed with 0.04 g of catalyst using a <u>Micromeritics AutoChem</u> <u>2910</u> Analyser Effluents gases were analysed after reaching a steady state by means of <u>Mass spectrometer</u> piloted with software (Quadstar, 32 Bits	2910

Sol–Gel Synthesis of MoO 3/SiO 2 Composite for Catalytic Application in	AP Amrute, A Bordoloi, N Lucas, K Palraj, SB Halligudi - Catalysis Letters, 2008	The total acidities of the catalysts were measured by temperature programmed desorption of NH 3 (NH 3 -TPD) using a <u>Micromeritics AutoChem-2910</u> instrument The products were identified by GC-MS (QP- 5000 <u>Mass spectrometer</u> , GC-17A Gas Chromatograph) analysis	2910
Stability of lanthanum oxide-based H2S sorbents in realistic fuel processor/fuel cell	I Valsamakis, R Si, M Flytzani- Stephanopoulos - Journal of Power Sources, 2009	The sorbent BET surface areas were measured by single-point N 2 adsorption/desorption cycles using a Micromeritics model AutoChem II 2920 concentrations of H 2 O, CO and CO 2 in the outlet gas were monitored on-line by a quadrupole <u>mass spectrometer</u> and the	2920
Steam reforming of liquid hydrocarbon fuels for micro-fuel cells. Pre- reforming of model jet fuels over supported metal	Zheng, J., Strohm, J.J., Song, C., Fuel Processing Technology, 89 (4), p.440-448, Apr 2008	an automated catalyst characterization unit (<u>Micromeritics AutoChem 2910</u>) to determine the metal dispersion and metalGC coupled with a Shimadzu QP-5000 quadrupole <u>mass spectrometer</u> (MS) detector. The GC is equipped with a capillary	2910
Structural changes of nano- Pt particles during thermal ageing: Support-induced	JP Dacquin, M Cabié, CR Henry, C Lancelot, C Journal of Catalysis, 2010	Hydrogen temperature-programmed reduction (H 2 -TPR) was carried out in a <u>Micromeritics</u> <u>AutoChem II 2920</u> apparatus (5 vol.% H 2 /Ar) analysed with a μGC Varian CP-4900 chromatograph fitted with two thermal conductivity detectors and a Balzer <u>mass spectrometer</u>	2920
Structure and Electronic Properties of Ca-Doped CeO~ 2 and Implications kth.se	S De Carolis, JL Pascual, LGM Pettersson, M Journal of Physical , 1999 - ict.kth.se	were evaluated with a <u>Micromeritics</u> TPD/ TPR <u>2900</u> Chemisorption Analyzer using a nearly stoichiometric reactant gas mixture containing 3.20% CO and 1.65% SO2 in He. The composition of outlet gas was analyzed with a Balzers <u>mass</u> <u>spectrometer</u> model QMG421C, and	2900
Study of catalytic behaviour and deactivation of vanadyl-aluminum binary	FM Bautista, JM Campelo, D Luna, J Luque, JM Chemical Engineering, 2006	TPR experiments were performed in a Micromeritics TPD/TPR 2900 analyser [6]. Samples of 50 mg were first treated in Ar at 100 °C for 10 °C/min (temperature range, 30–1000 °C). The CO 2 signal was analysed with an on-line quadrupole <u>mass spectrometer</u> (Pfeiffer Vacuum	2900
STUDY OF N2O DECOMPOSITION OVER FE-ZSM-5 WITH TRANSIENTepfl.ch	P PREChTL - biblion.epfl.ch	Micromeritics AutoChem 2910 Analyzer 58 Potential IPCC Intergovernmental Panel on Climate Change IR Infrared MCT Mercury Cadmium Telluride MFI Crystalline structure of ZSM-5 zeolite (Mobil Five) MMO Methane Monooxygenase MS Mass spectrometer NSCR Non	2910
Study of the origin of the deactivation of a Pt/CeO2 catalyst during reverse water	A Goguet, F Meunier, JP Breen, R Burch, MI Petch, A Journal of Catalysis, 2004	2010) and was found to be 180 m 2 g -1 . The Pt dispersion was measured by H 2 chemisorption (<u>Micromeritics AutoChem 2910</u>) and 5 °C min -1 . The CO and CO 2 concentrations were followed at the outlet of the reactor with a quadrupole <u>mass spectrometer</u> (VG Gaslab300	2910
Study of the temperature- programmed oxidative degradation of hydrocarbons over	JI Gutiérrez-Ortiz, B de Rivas, R López- Fonseca Journal of thermal, 2007	investigated. The effluent stream was analysed by a TCD coupled to a <u>mass spectrometer</u> (HAL Quadropole <u>Mass spectrometer</u> , Hyden Analytical). The experiments were conducted on a <u>Micromeritics AutoChem</u> 2910 instrument	2910
Study on Copper-based Catalysts for Synthesis of N, N'-bis (1, 4- dimethylpentyl)	ZD Pan, YJ Ding, L Yan, XM Li, GP Jiao, HY Luo - Catalysis Letters, 2008	monitoring simultaneously the signal of H2 (m/z = 2) with a <u>mass spectrometer</u> detector (Prisma). H2 temperature programmed reduction (H2-TPR) measurements were performed on a <u>Micromeritics</u> <u>AutoChem</u> <u>2910</u> apparatus	2910

Study on Ni-Re-K/Al2O3 catalysts for synthesis of N, N'-di-sec-butyl p-phenylene 	Z Pan, Y Ding, D Jiang, X Li, G Jiao, H Luo - Applied Catalysis A, General, 2007	Temperature programmed desorption of ammonia (NH 3 -TPD) was performed in a <u>Micromeritics</u> <u>AutoChem</u> <u>2910</u> a ramp of 10 K/min in a helium flow of 40 ml/min and the NH 3 (m/z = 16) was monitored simultaneously with a quadrupole <u>mass spectrometer</u> detector (Balzers	2910
Study on the Ti and Al coincorporation into the MFI zeolitic structurersc.org	G Ovejero, R Grieken, MA Uguina, DP Serrano, Journal of Materials , 1998	1 and 2. carried out with a <u>Micromeritics</u> TPD/TPR <u>2900</u> apparatus. The sample was first outgassed by thermal treatment, from Samples 4–12 Person et al.22 to synthesize silicalite-1. This method allows to continuously using a quadrupole <u>mass spectrometer</u> (Hiden	2900
Sulphur poisoning of palladium catalysts used for methane combustion: Effect of	LS Escandón, S Ordóñez, A Vega, FV Díez - Journal of Hazardous, 2008	Temperature-programmed desorption analyses were carried out using a <u>Micromeritics</u> TPD/TPR <u>2900</u> Analyzer, operating at ambient pressure room temperature to 900 °C at 10 °C min −1 , and the gas leaving the reactor was analysed by a <u>mass spectrometer</u> (GASLAB 300	2900
Sulphur poisoning of transition metal oxides used as catalysts for methane	S Ordóñez, JR Paredes, FV Díez - Applied Catalysis A, General, 2008	Temperature-programmed desorption was carried out in a <u>Micromeritics</u> TPD/TPR <u>2900</u> apparatus connected to a TCD or a MS detector (Gaslab-300). Samples of 10 mg of catalysts were heated from 50 to 1000 °C at 10 °C/min in a flow of 0.1 L/min of pure He	2900
Supported bimetallic AuRh/±c-Al2O3 nanocatalyst for the selective catalytic reduction of NO by propylene	Liu, L., Guan, X., Li, Z., Zi, X., Dai, H., He, H., Applied Catalysis B, Environmental, 90 (1), p.1-9, Jul 2009	2020 apparatus from Micromeritics. The samples were02-H2titration on AutoChem 2920 II chemical adsorption apparatus from Micromeritics. The 0.2g of catalystcatalysts was performed on AutoChem 2920 II chemical adsorptionon-line by a quadrupole mass spectrometer (HIDEN HPR20 equipped	2920
Supported bimetallic AuRh/γ-Al2O3 nanocatalyst for the selective catalytic	L Liu, X Guan, Z Li, X Zi, H Dai, H He - Applied Catalysis B, Environmental, 2009	H 2 -TPR of catalysts was performed on <u>AutoChem 2920</u> II chemical adsorption apparatus from <u>Micromeritics</u> by using z = 32), N 2 (m/z = 28), N 2 O (m/z = 44) and NO 2 (m/z = 46) in NO-TPD process were monitored on-line by a quadrupole <u>mass spectrometer</u> (HIDEN HPR20	2920
Supported metal particles from LDH nanocomposite precursors: control of the	C Gerardin, D Kostadinova, N Sanson, B Coq, D Tichit - Chem. Mater, 2005	K in a stream of He or synthetic air (flow: 20 cm 3 min - 1). A Balzers QMS 421 quadrupole <u>mass</u> <u>spectrometer</u> equipped with The H 2 temperature-programmed reduction (TPR) analyses were done using a <u>Micromeritics AutoChem</u> <u>2910</u> instrument with a TCD detection (m = 30	2910
Surface characterization of supported Pd catalysts activated with chiral hydrogen	MA Aramendia, Y Aviles, V Borau, C Jimenez, JM Langmuir, 1999	TPD-TCD Experiments. Temperature-programmed desorption experiments were carried out on a Micromeritics TPD/TPR 2900 instrument The oven outlet was connected to the <u>mass spectrometer</u> probe to collect samples in a continuous fashion	2900
Surface characterization studies of TiO2 supported manganese oxide catalysts foruc.edu	PR Ettireddy, N Ettireddy, S Mamedov, P Applied Catalysis B, , 2007	The oxygen uptake was quantified by a TCD connected to a 2910 AutoChem I (Micromeritics instrument) The reactants and products were analyzed on-line using a Quadrapole <u>mass</u> <u>spectrometer</u> (MKS PPT-RGA), and a NO x analyzer (Eco Physics CLD 70S)	2910
Surface properties and catalytic performance for ethane combustion of La1	YN Lee, RM Lago, JLG Fierro, V Cortés, F Applied Catalysis A, , 2001	Temperature programmed reduction (TPR) and O 2 temperature programmed desorption (TPO) profiles were obtained in a <u>Micromeritics 2900</u> instrument at 10 K min -1 and the desorption products O 2, CO 2 and H 2 O monitored by a <u>mass spectrometer</u> detector Balzers QMG	2900

Surface reconstruction of supported Pd on LaCoO3: Consequences on the catalytic properties in the decomposition of N2O	Dacquin, J.P., Dujardin, C., Granger, P., Journal of Catalysis, 253 (1), p.37-49, Jan 2008	Temperature-programmed reduction (TPR) was carried out in a <u>Micromeritics AutoChem II 2920</u> (5 vol% H2/Ar). In situ X-ray diffractiontwo thermal conductivity detectors and a Balzer <u>mass spectrometer</u> for the detection and the quantification of O2and	2920
SURVIVAL OF LIGNIN- DERIVED STRUCTURAL UNITS IN ANCIENT COALIFIED WOOD SAMPLES.	6-Mar	data not available Pyrolysis-gas chromatography- mass spectrometry was performed on a Dupont 4908 gas chromatograph- <u>mass spectrometer</u> system interfaced with a Technivent Vector 1 data system and a Chemical Data Systems model 120 pyroprobe. Pyrolysis	
Synthesis and characterization of MgO-B 2 O 3 mixed oxides prepared byrsc.org	MA Aramendía, V Boráu, C Jiménez, JM Marinas Journal of Materials , 1999	the boron atoms are considered located in a trigonal or <u>Micromeritics</u> TPD/TPR <u>2900</u> —VG Sensorlab quadrupole tetragonal environment within the magnesium network. As we <u>mass spectrometer</u> . The optimum TPD conditions were as	2900
Synthesis and characterization of ZrO 2 as an acid-base catalystrsc.org	MA Aramendía, V Boráu, C Jiménez, JM Marinas Journal of the Chemical, 1997	TPD–MS experiments TPD»MS experiments were carried out on a <u>Micromeritics</u> TPD/TPR <u>2900</u> instrument on line with a VG Sensorlab quadrupole <u>mass spectrometer</u> . The optimum TPD conditions were as follows: heating	2900
Synthesis and properties of PdSn/Al2O3 and PdSn/SiO2 prepared by solvated metal atom dispersed method	Cardenas, G., Oliva, R., Reyes, P., Rivas, B.L., Journal of Molecular Catalysis A: Chemical, 191 (1), p.75-86, Jan 2003	343 K was carried by a pulse method in a TPD/TPR 2900 Micromeritics system provided with a thermal conductivity detectorreactants and products were carried out by on line mass spectrometer HIDEN HAL 200. The reactor was kept in a furnace	2900
Synthesis of alkoxide- derived V-Nb catalysts prepared by sol–gel route	M Catauro, C Pagliuca, L Lisi, G Ruoppolo - Thermochimica Acta, 2002	was employed for the BET measurement of surface areas by N 2 adsorption at 77 K. Temperature-programmed reduction (TPR) was carried out using of a <u>Micromeritics TPD/TPR 2900</u> analyzer equipped with a TCD and coupled with a Hiden HPR 20 <u>mass spectrometer</u>	2900
Synthesis of Biodiesel via Deoxygenation of Stearic Acid over Supported Pd/C 	S Lestari, I Simakova, A Tokarev, P Mäki- Arvela, K Catalysis Letters, 2008	The metal dispersion was determined by CO pulse chemisorption using <u>AutoChem</u> 1900 (<u>Micromeritics</u>) The product identification was validated with a gas chromatograph— <u>mass spectrometer</u> (GC–MS). 248 S. Lestari et al. 123 Page 3. 2.3.2 Gas Phase Analysis	2900
Synthesis of camphene from α-pinene using SO32-functionalized MCM-41 as	M Roman-Aguirre, YP Gochi, AR Sanchez, L de Applied Catalysis A, , 2007	The acid sites per gram of each catalyst were quantified by thermal desorption of NH 3 in a Micromeritics AutoChem 2910 analyzer present in the samples were identified by analysis in a Thermofinigan GC Top 8000/Voyager MS gas chromatograph/mass spectrometer system	2910
Synthesis of gasoline-range hydrocarbons over Mo/HZSM-5 catalysts	S Liu, AC Gujar, P Thomas, H Toghiani, MG Applied Catalysis A,, 2009	Catalyst characterization. Temperature programmed reduction/reaction (TPR) of 10%Mo/HZSM-5 was conducted with a Micromeritics AutoChem 2910 apparatus combined with a Dycor Dymaxion Quadrapole Mass spectrometer from Ametek Process Instruments	2910
Synthesis of LaFeO 3 catalytic materials and their sensing propertiesscichina.com	SL Bai, BJ Shi, LJ Ma, PC Yang, ZY Liu, DQ Li, Science in China, 2009	sensing materi- als' surface, TPD experiments were performed on La- FeO3 and 4% MgO-coated LaFeO3 nanocomposites prepared by citric acid method by a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyzer equipped with a TC detector and coupled with a HPR 20 <u>mass spectrometer</u>	2900

Synthesis of reactive nano- Fe/Pd bimetallic system- impregnated activated carbon	H Choi, SR Al-Abed, S Agarwal, DD Dionysiou - Chem. Mater, 2008	H 2 -temperature programmed reduction (H 2 - TPR) technique was adopted on AutoChem 2910 TPD/TPR instrument (Micromeritics) to investigate D-8 naphthalene in dichloromethane), and analyzed in a gas chromatograph (GC, HP 6890)/ <u>mass spectrometer</u> (MS, HP	2910
Synthesis of ZSM-5 zeolite in fluoride media: an innovative approach to tailor both crystal size and acidity	Louis, B., Kiwi- Minsker, L., Microporous and Mesoporous Materials, 74 (1), p.171-178, Sep 2004	properties. The measurements were performed in a <u>Micromeritics AutoChem 2910</u> analyser provided with a quartz plug-flow reactor. A ThermoStar 200 (Pfeiffer Vacuum) mass-spectrometer was used to analyse the gas-phase composition	2910
Synthesis, characterization and activation of quaternary layered double hydroxides	NN Das, R Das - Reaction Kinetics, Mechanisms and Catalysis	The gases evolved during decomposition were analyzed by an online Leybold Transpector SQX quadrupole <u>mass spectrometer</u> C. Temperature programmed reduction (TPR) of the calcined samples were performed in a <u>AutoChem</u> <u>2910</u> from <u>Micromeritics</u> as described	2910
TAP reactor study of the deep oxidation of propane using cobalt oxide and gold	B Solsona, T García, GJ Hutchings, SH Taylor, Applied Catalysis A,, 2009	Temperature-programmed reduction was carried out in a <u>Micromeritics AutoChem 2910</u> equipped with a TCD detector recorded at the reactor outlet by three PC interfaced Balzers QMA124 quadrupole <u>mass spectrometers</u> (QMS) in line; a fourth <u>mass spectrometer</u> is positioned	2910
Temperature-programmed desorption study of molecular oxygen adsorbed on MFI	MH Kim, SJ Kim, SB Hong, G Seo, YS Uh - Korean Journal of Chemical, 1998	Temperature-programmed desorption (TPD) of 02 was re- corded on a Micromeritics TPD/TPR 2900 analyzer 373 to 873 K with heating rates ranging from 5 to 30 K. In all TPD experiments the desorbed gas was identified as O2 alone by a Balzers MSC 200 <u>mass spectrometer</u>	2900
Temperature-programmed desorption study of Re/±c-Al2O3 catalysts prepared from Re2(CO)10 precursor	Raty, J., Pakkanen, T.A., Applied Catalysis A: General, 208 (1), p.169-175, Feb 2001	TPD) were performed with a <u>Micromeritics</u> <u>AutoChem 2910</u> analyser. A thermalinstrument to a quadrupole <u>mass spectrometer</u> (HP 5920). A typical TPDcatalyst (8.9 wt.%). shows the <u>mass spectrometer</u> signal from the catalyst	2910
The alteration of the performance of field-aged Pd-based TWCs towards CO and C3H6 oxidation	Heo, I., Choung, J.W., Kim, P.S., Nam, I.S., Song, Y.I., In, C.B., Yeo, G.K., Applied Catalysis B, Environmental, 92 (1), p.114-125, Oct 2009	volumetric methods (Micromeritics, ASAP 2010chemisorption method (AutoChem II 2920, Micromeritics). Details areH2/Ar with AutoChem II 2920. H2temperaturePackard) and a mass spectrometer (QMI422, Pfeiffer	2920
The catalytic reforming of bio-ethanol over SiO2 supported ZnO catalysts: The role	E Seker - International Journal of Hydrogen Energy, 2008	source Hg or less. In addition, CO 2 TPD was performed to measure the basicity of all catalysts using <u>Micromeritics AutoChem 2910</u> equipped with an inline Balzers Thermostar GS300 quadrupole <u>mass spectrometer</u> . 0.1 g of a	2910
The effect of metal ions in MNaY-zeolites for the adsorptive removal of	YH Kim, HC Woo, D Lee, HC Lee, ED Park - Korean Journal of Chemical , 2009	desorption (TPD) of THT adsorbed on the adsorbents was conducted over 30 mg of sample at a heating rate of 10 K/min monitoring thermal conductivity detector (TCD) and mass signals using an <u>AutoChem</u> <u>2910</u> unit (<u>Micromeritics</u>) and a <u>mass</u> <u>spectrometer</u> (QMS 200	2910

The interaction of N2O with ZSM-5-type zeolites: A transient, multipulse	A Ates, A Reitzmann - Journal of Catalysis, 2005	All transient-response studies were performed in an apparatus containing a quartz glass reactor (AutoChem 2910, Micromeritics). A quadruple mass spectrometer (QMS 422; Pfeiffer Vacuum) was used in an on-line mode to determine the composition of the gas phase	2910
The promoted effect of UV irradiation on preferential oxidation of CO in an H2-rich stream over Au/TiO2	W Dai, X Zheng, H Yang, X Chen, X Wang, P Liu, Journal of Power, 2009	However, no other product in outlet was observed except that H 2 O and CO 2 in all reaction systems. 2.4. Temperature program desorption (TPD). Temperature program desorption (TPD) of Au/TiO 2 was tested in Micromeritics AutoChem 2910 instrument	2910
The promotion effects of Mn, Li and Fe on the selectivity for the synthesis of C2	HM Yin, YJ Ding, HY Luo, WM Chen, Natural gas conversion, 2004	CO-TPD experiments were performed on Micromeritics AutoChem 2910. After CO adsorption saturation, the catalyst bed was swept with He for 5 min. Each experiment was performed in a He flow (flow rate= 20 ml/min) with a quadruple mass spectrometer (QMS, Balzers	2910
The quality of SiO2- coatings on flame-made TiO2-based nanoparticles135.196.210.1 95	A Teleki, MK Akhtarb, SE Pratsinis - matrix - 135.196.210.195	The chemisorption of isopropanol on FSP-made particles was investigated on a <u>Micromeritics</u> <u>AutoChem II 2920</u> unit The off-gases from the <u>AutoChem</u> were analyzed by a <u>mass spectrometer</u> (MS; Thermo Star, Pfeiffer Vacuum, SEM and emission mode)	2920
The reactivity of ruthenium mono (bipyridine) carbonyl complexes in an alcoholic	S Luukkanen, M Haukka, O Laine, T Venäläinen, Inorganica Chimica , 2002	TPD profiles of the ruthenium bipyridine complexes were recorded on a Micromeritics TPD/TPR 2900 instrument under a He-flow of 10 ml min The EI ionisation mass spectra were recorded with a JEOL JMS D300 magnetic sector mass spectrometer and the LD ionisation mass	2900
The reactivity of ruthenium mono(bipyridine) carbonyl complexes in an alcoholic solution of alkali metal carbonates	Luukkanen, S., Haukka, M., Laine, O., Venalainen, T., Vainiotalo, P., Pakkanen, T.A., Inorganica Chimica Acta, 332 (1), p.25- 29, Apr 2002	were recorded on a Micromeritics TPD/TPR 2900 instrument undermagnetic sector mass spectrometer and the LD ionisationtime-of-flight mass spectrometer. Polymeric samplesintroduced into the mass spectrometer under an argon	2900
The Use of a Jet-Stirred Continuously Stirred Tank Reactor (CSTR) to Study the	GA Foulds, BG Charlton, BT Leb, JC gas conversion four, 1997	A portion of the exit gas from the reactor (15ml/min NTP) was diverted to a quadropole <u>mass spectrometer</u> to measure the concentration of the TPR analysis were carried out in a <u>Micromeritics</u> TPD/TPR <u>2900</u> apparatus, using~ 50mg of the sample in a 10% H2/Ar flow at 10 C/min	2900
Thermally reduced ruthenium nanoparticles as a highly activenus.edu.sg	F Su, L Lv, FY Lee, T Liu, Al Cooper, XS Zhao - Science, 2007 - nus.edu.sg	an inductive-coupled plasma atomic <u>mass</u> <u>spectrometer</u> (ICP-MS) on a Perkin-Elmer ELAN6100 at a wavelength of 100.9 nm.20 H2 chemisorption on Ru catalysts was attempted to measure using various methods such as <u>Micromeritics</u> <u>AutoChem II</u> <u>2920</u> automated catalyst	2920
TiO2 supported vanadyl phosphate as catalyst for oxidative dehydrogenation of	P Ciambelli, P Galli, L Lisi, MA Massucci, P Applied Catalysis A,, 2000	Temperature Programmed Reduction (TPR) with hydrogen and Temperature Programmed Desorption (TPD) of NH 3 were carried out using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyser equipped with a TC detector and coupled with a Hiden HPR 20 <u>mass spectrometer</u>	2900

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Titania supported bimetallic transition metal oxides for low-temperature SCR of NO 	PM Sreekanth, DA Pena, PG Smirniotis - Ind. Eng. Chem. Res, 2006	The temperature-programmed reduction (H 2 - TPR) experiments were carried out from 353 to 1223 K on a <u>Micromeritics AutoChem 2910</u> instrument in situ by passing oxygen for 2 h at 673 K. The products were analyzed online using a Quadrapole <u>mass spectrometer</u> (MKS PPT	2910
Titanium oxide loaded zeolites as photocatalysts for the cyclization of	KV Subba Rao, B Srinivas, M Subrahmanyam - Catalysis Letters, 2003	Temperature-programmed desorption (TPD) of ammonia experiments were carried out on an Auto-Chem 2910 (Micromeritics, USA) instrument Low-resolution EI mass spectra were recorded on a VG 7070H Micromass mass spectrometer at 473K, 70eV, with a trap current	2910
Total oxidation of chlorinated VOCs on supported oxide catalysts	Bertinchamps, Fabrice , Nov 2005	primary ion beam to desorb and ionize species from a sample surface. The resulting secondary ions are accelerated into a <u>mass spectrometer</u> , where they are mass analyzed by measuring their time-of-flight from the sample surface to the detector. An image is generated	
Total oxidation of toluene over calcined trimetallic hydrotalcites type catalysts	LA Palacio, J Velásquez, A Echavarría, A Faro, Journal of Hazardous, 2009	Temperature-programmed reduction with hydrogen was carried out in a <u>AutoChem II</u> <u>Micromeritics</u> equipment in the literature from identification of the gases evolved during the thermal decomposition process continuously monitored with a <u>mass spectrometer</u> [19] and [20]	2920
Transient multi pulse method for the determination of N 2 O-interaction with ZSM-5	A Ates, A Reitzmann - Reaction Kinetics and Catalysis Letters, 2005	All catalytic experiments were performed in a fully-automated apparatus (<u>Micromeritics</u> , <u>AutoChem</u> <u>2910</u>) containing a quartz glass reactor operated in the plug-flow mode. A quadruple <u>mass spectrometer</u> (QMS 422, Pfeiffer Vacuum, Germany) was used to determine the	2910
Transient response method for characterization of active sites in HZSM-5 with lowepfl.ch	L Kiwi-Minsker, DA Bulushev, A Renken - Catalysis Today, 2004	The active sites concentration measurements, reactivity and temperature-programmed (TPD) studies were performed in a <u>Micromeritics</u> <u>AutoChem 2910</u> analyzer provided with a quartz plug-flow reactor. A ThermoStar 200 (Pfeiffer Vacuum) mass-spectrometer was used to	2910
Use of test reactions for the characterisation of bimetallic Pt-Sn/Al2O3 catalysts	MP González- Marcos, B Iñarra, JM Guil, MA Applied Catalysis A,, 2004	by temperature-programmed desorption (TPD) of ammonia in a Micromeritics AutoChem 2910 , equipped dispersion was evaluated by H 2 chemisorption in a Micromeritics ASAP 2010 detectors were: an AED, used for quantification, and a <u>mass spectrometer</u> , for identification of	2910
Utilization of High Specific Surface Area CuO- CeO2 Catalysts for High	P Djinović, J Batista, B Ćehić, A Pintar - 2009	with CO 2 were performed at atmospheric pressure using a <u>Micromeritics AutoChem II 2920</u> test tube, which was inserted into an electric furnace of <u>AutoChem II 2920</u> reactants during activity probing tests were simultaneously analyzed by <u>mass spectrometer</u> (Pfeiffer Vacuum	2920
Utilizing full-exchange capacity of zeolites by alkaline leaching: Preparation of Fe	I Melian-Cabrera, S Espinosa, JC Groen, B v/d Journal of Catalysis, 2006	Temperature-programmed reduction (TPR) with H 2 was performed in a <u>Micromeritics</u> TPD/TPR <u>2900</u> apparatus, using a high-purity mixture of 5 vol% H were carried out in a Mettler Toledo system (TGA/SDTA 851E) equipped with a Thermo-Star quadrupole <u>mass spectrometer</u>	2900

Vanadium- metal(IV)phosphates as catalysts for the oxidative dehydrogenation of ethane	Lisi, L., Ruoppolo, G., Casaletto, M.P., Galli, P., Massucci, M.A., Patrono, P., Pinzari, F., Journal of Molecular Catalysis. A, Chemical, 232 (1), p.127-134, May 2005	flow. Temperature programmed reduction (TPR) was carried out using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyser equipped with a TCD and coupled with a Hiden HPR 20 <u>mass spectrometer</u> . The sample (100mg) was reduced by a 2% H2/Ar mixture (25cm3min-1	2900
Vanadyl phosphate dihydrate supported on oxides for the catalytic conversion of	L Lisi, P Patrono, G Ruoppolo - Journal of Molecular Catalysis. A,, 2003	Temperature programmed reduction (TPR) with hydrogen were carried out using a <u>Micromeritics</u> TPD/TPR <u>2900</u> analyser equipped with a TC detector and coupled with a Hiden HPR 20 <u>mass spectrometer</u> reducing the sample with a 2% H 2 /Ar mixture (25 cm 3 min -1) and	2900
Vapor-phase hydrogenolysis of biomass- derived lactate to 1, 2- propanediol over	L Huang, Y Zhu, H Zheng, M Du, Y Li - Applied Catalysis A, General, 2008	The amount of chemisorbed hydrogen was measured using the <u>Micromeritics</u> Auto Chem. <u>2920</u> by temperature program desorption (TPD) technique The liquid products were identified by gas-chromatograph (6890N, Agilent) with <u>mass spectrometer</u> (5973, Agilent)	2920
Water Vapor Effects in N2O Decomposition over Fe- ZSM-5 Catalysts with Low 	DA Bulushev, PM Prechtl, A Renken, L Kiwi Ind. Eng. Chem, 2007	Catalytic Activity Measurements. Catalytic activity measurements and temperature- programmed desorption studies were carried out in a <u>Micromeritics AutoChem 2910</u> analyzer. A ThermoStar 200 (Pfeiffer Vacuum) quadrupole <u>mass spectrometer</u> was used for gas analysis	2910
Water-gas shift reaction over supported Pt-CeOx catalysts	Kim, Y.T., Park, E.D., Lee, H.C., Lee, D., Lee, K.H., Applied Catalysis B, Environmental, 90 (1), p.45-54, Jul 2009	conducted in an AutoChem 2910 unit (Micromeritics) equipped withand an on-line mass spectrometer (QMS 200, Pfeiffercarried out in an AutoChem 2910 unit (Micromeriticsand an on-line mass spectrometer (QMS 200, Pfeiffer	2910
XPS study of spent FCC catalyst regenerated under different conditions	RE Roncolatto, MJB Cardoso, HS Cerqueira, YL Ind. Eng. Chem, 2007	by the American Petroleum Institute), containing 1800 ppm of basic N and 3000 ppm of total N. TPO (temperature-programmed oxidation) was performed in a <u>Micromeritics AutoChem 2910</u> equipment connected to a <u>mass spectrometer</u> Balzers Thermo Star 422 by a heated tube	2910