

**Brønsted base (Brönsted base)**

A *molecular entity* capable of accepting a *hydron* (proton) from an acid (i.e. a ‘hydron acceptor’) or the corresponding *chemical species*. For example:  $\text{OH}^-$ ,  $\text{H}_2\text{O}$ ,  $\text{CH}_3\text{CO}_2^-$ ,  $\text{HSO}_4^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Cl}^-$ .

See also *conjugate acid–base pair*.

1994, 66, 1091